Lecture 11 Tue. 2.23.2016

(Ch. 8 of Unit 1)

Geometry of common power-law potentials

Geometric (Power) series

"Zig-Zag" exponential geometry

Projective or perspective geometry

Parabolic geometry of harmonic oscillator kr²/2 potential and -kr¹ force fields

Coulomb geometry of -1/r-potential and -1/r²-force fields

Compare mks units of Coulomb Electrostatic vs. Gravity

Geometry of idealized "Sophomore-physics Earth"

Coulomb field <u>outside</u> Isotropic Harmonic Oscillator (IHO) field <u>inside</u>

Contact-geometry of potential curve(s)

"Crushed-Earth" models: 3 key energy "steps" and 4 key energy "levels"

Earth matter vs nuclear matter:

Introducing the "neutron starlet" and "Black-Hole-Earth"

Introducing 2D IHO orbits and phasor geometry

Phasor "clock" geometry

Geometry of common power-law potentials

Geometric (Power) series

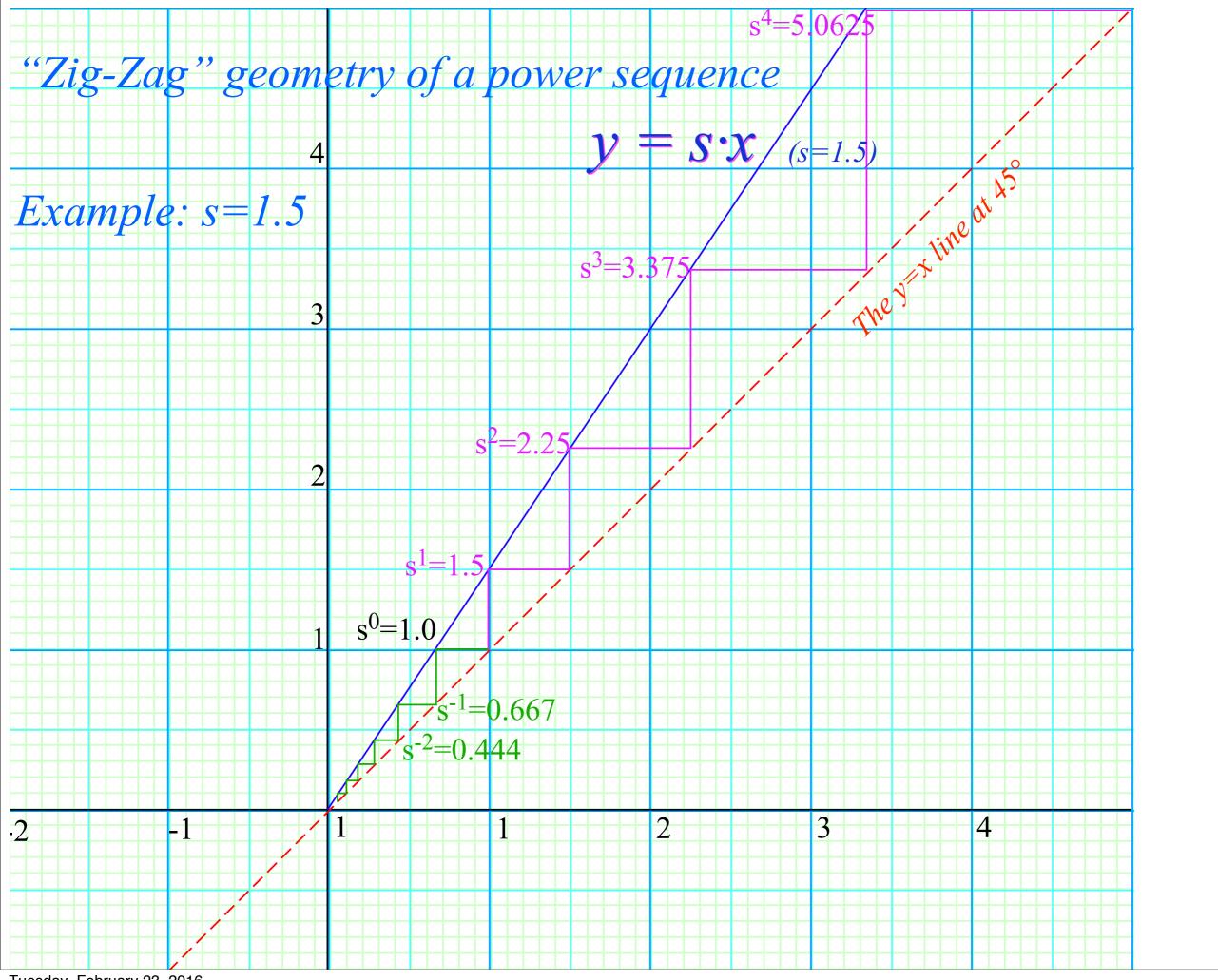
"Zig-Zag" exponential geometry

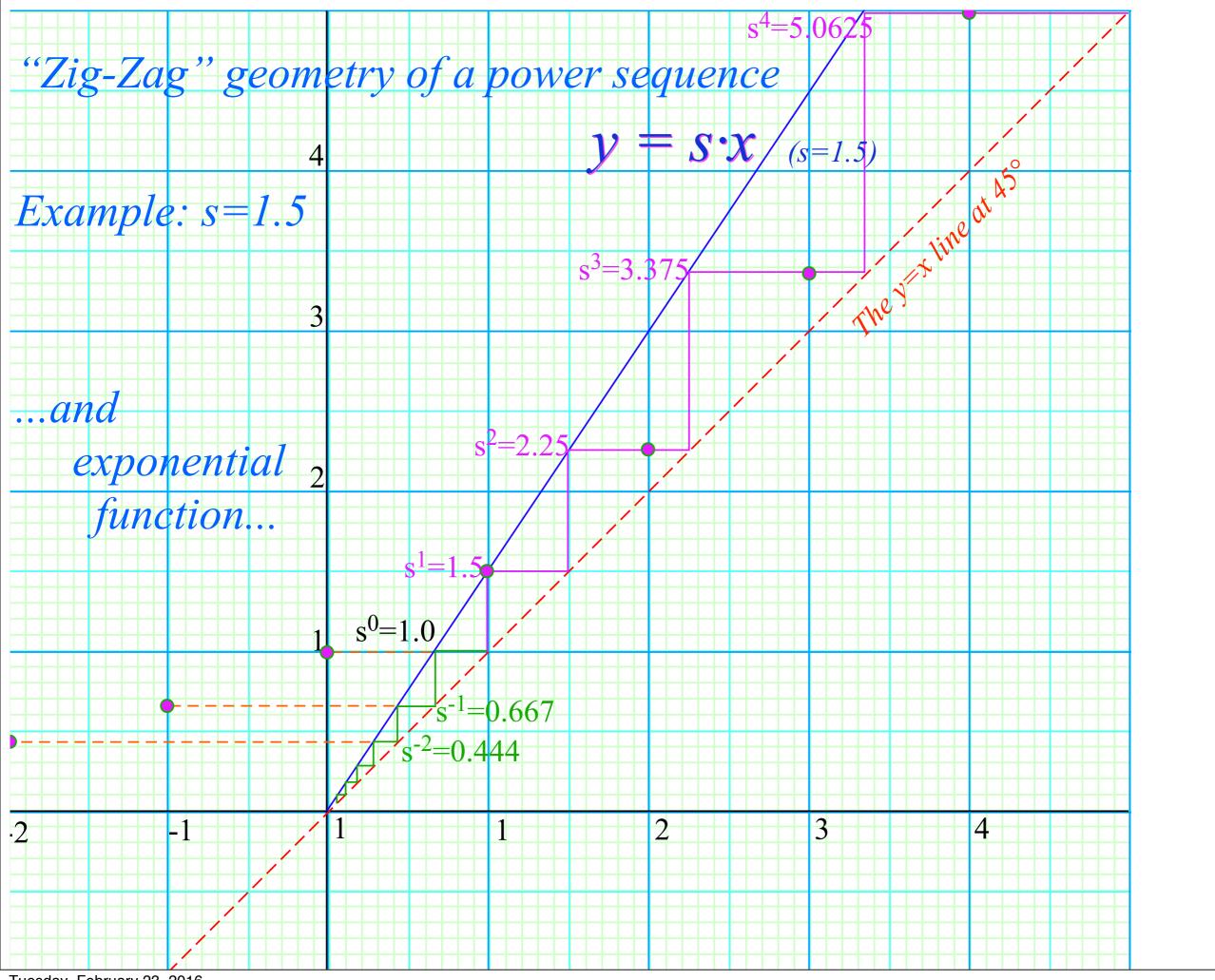
Projective or perspective geometry

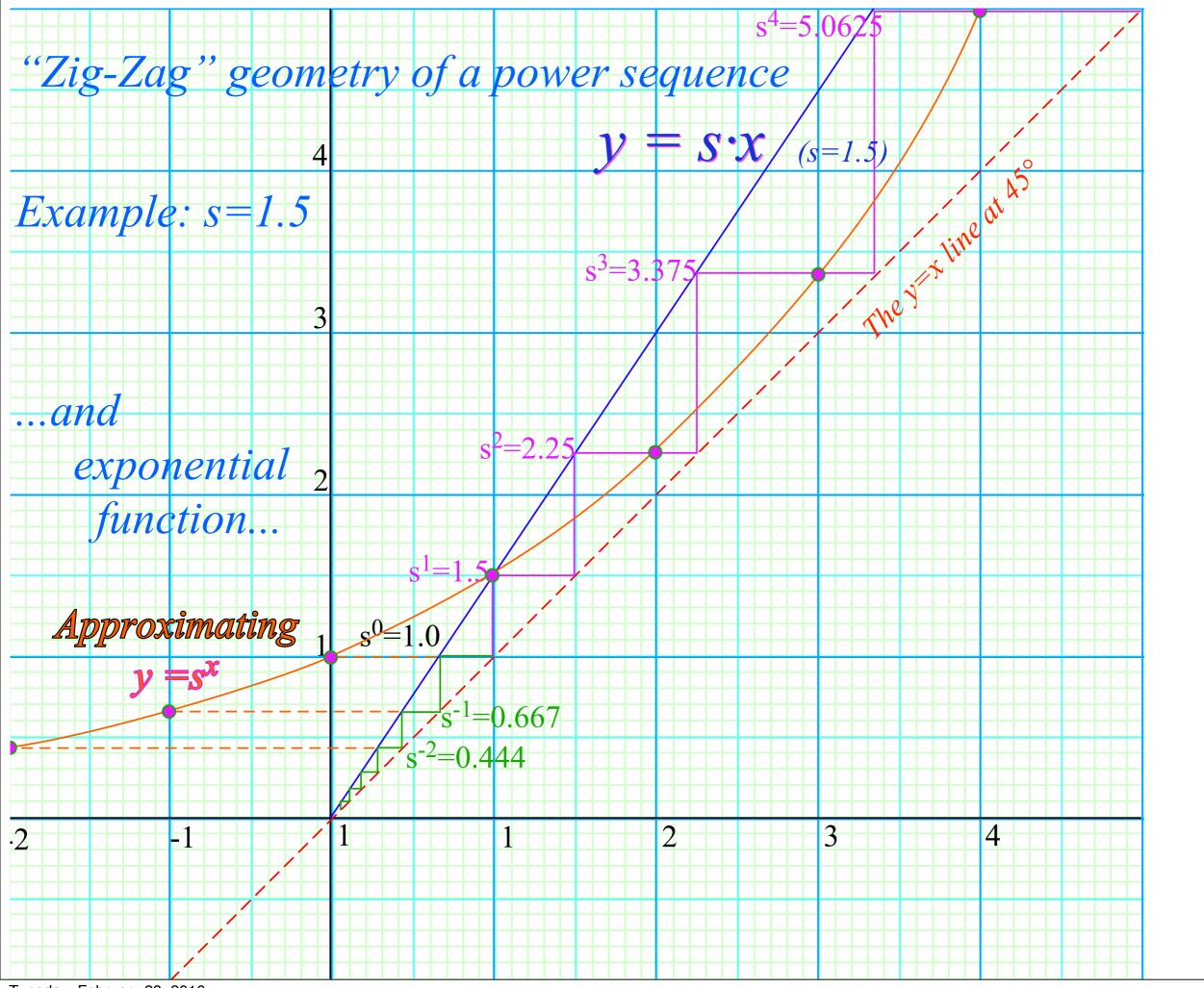
Parabolic geometry of harmonic oscillator kr²/2 potential and -kr¹ force fields

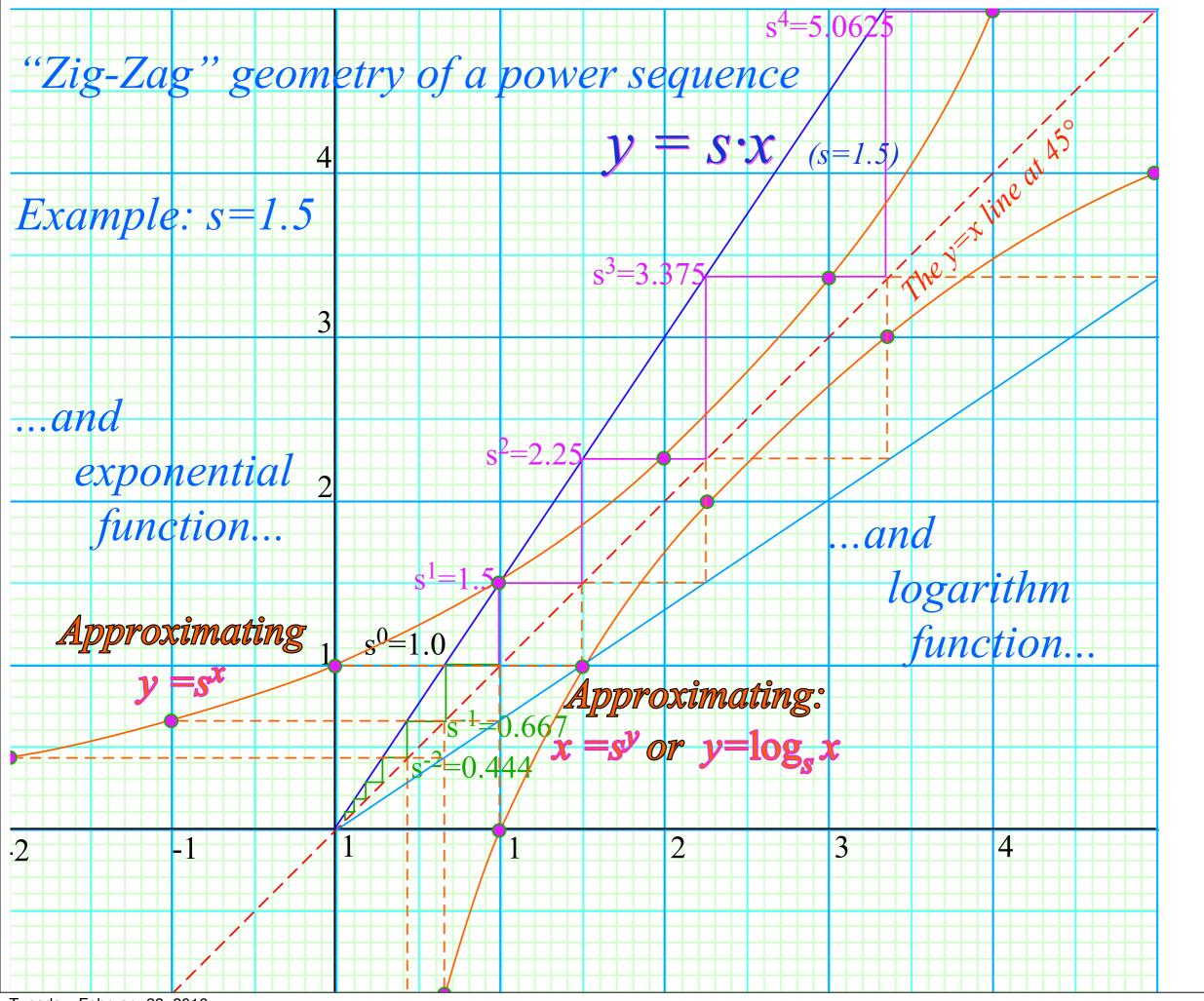
Coulomb geometry of -1/r-potential and -1/r²-force fields

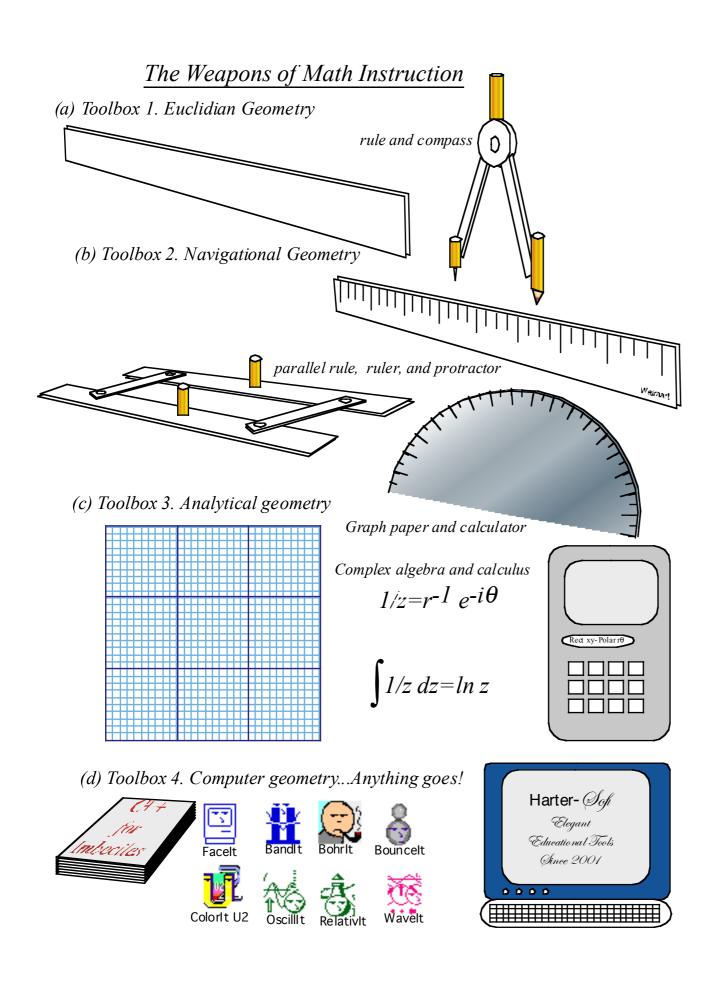
Compare mks units of Coulomb Electrostatic vs. Gravity







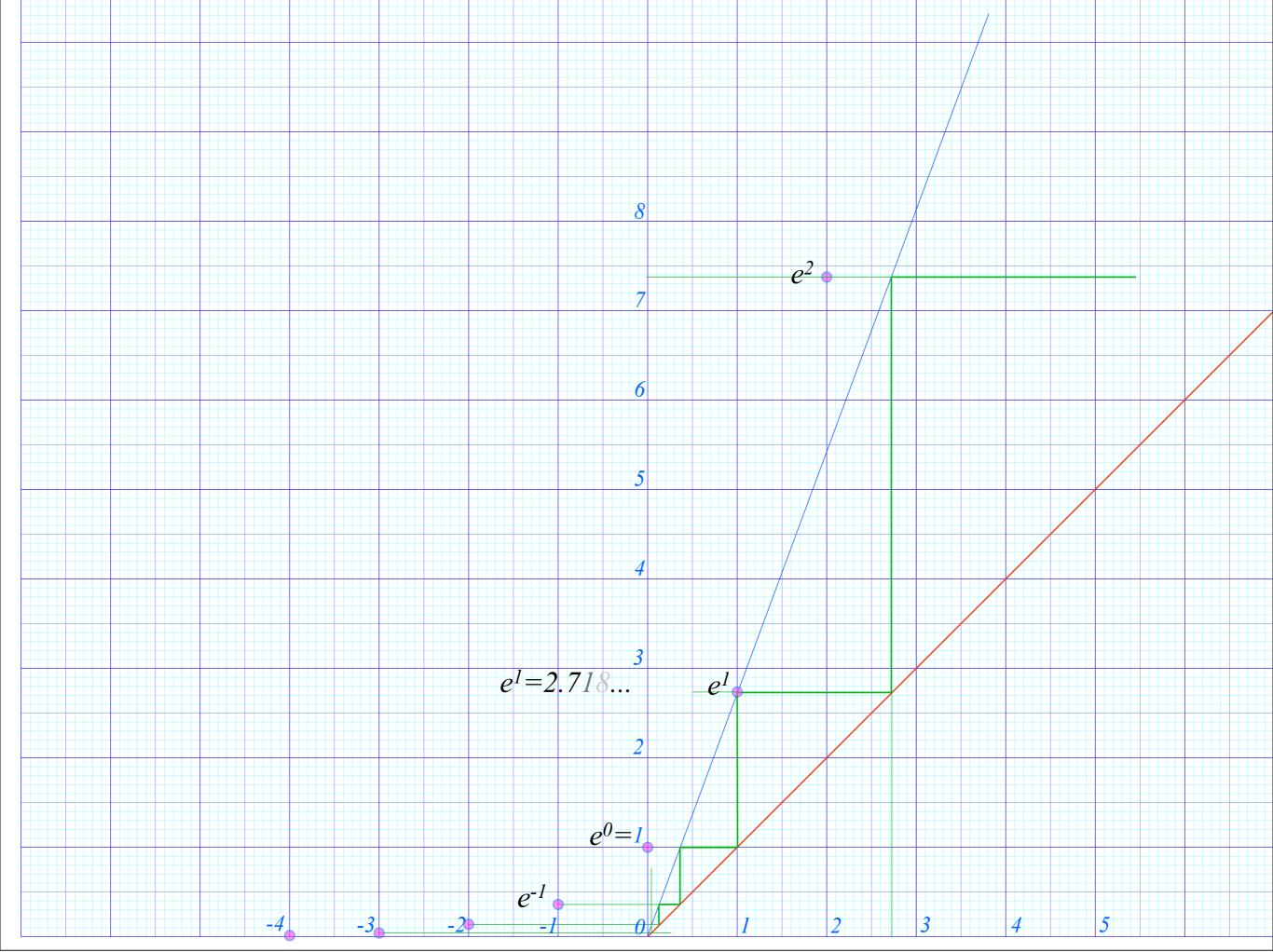


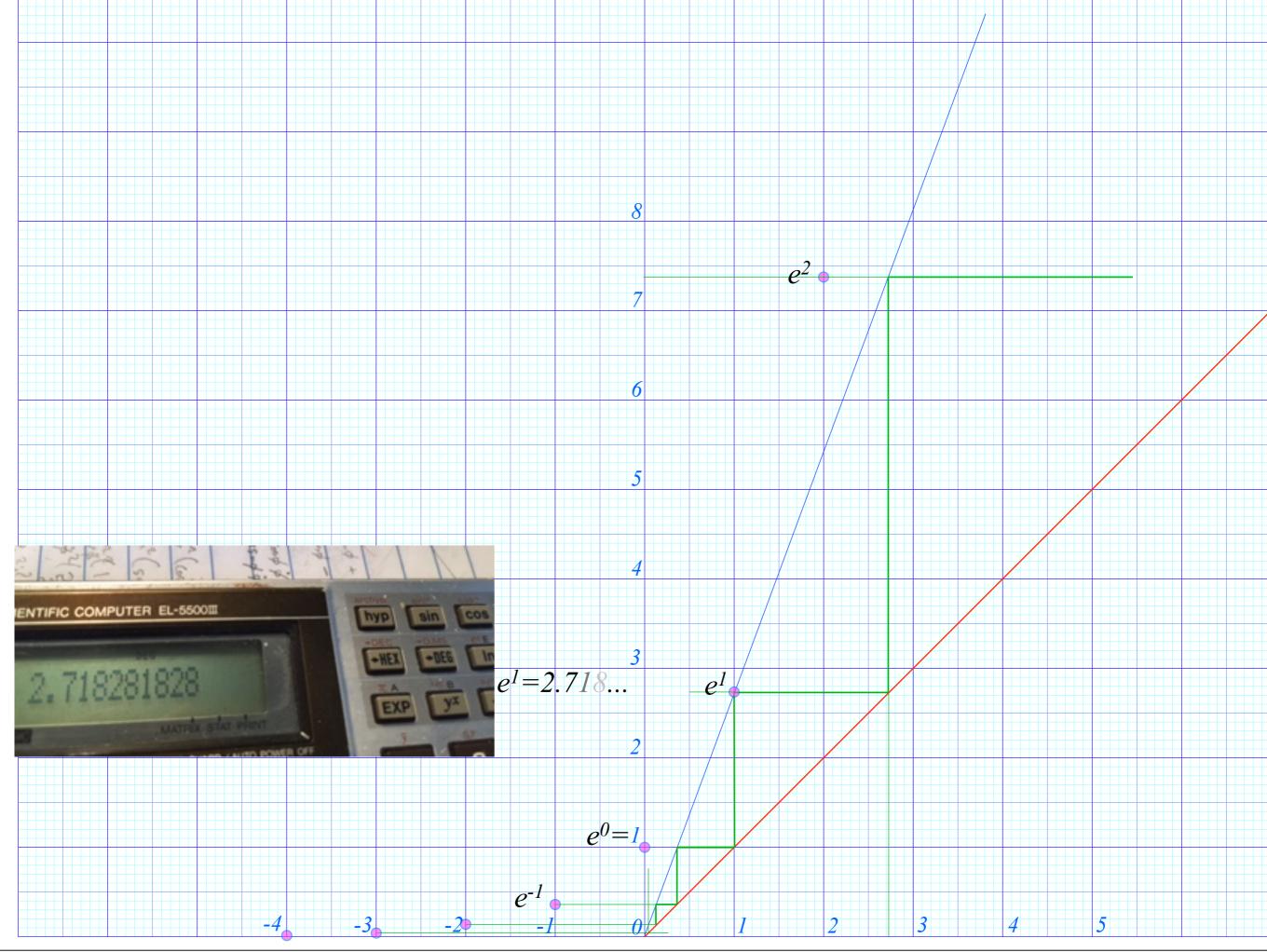


So far we mostly use Toolbox (a-b)

What follows uses
Toolbox (c) ...

...and Toolbox (d)





Geometry of common power-law potentials

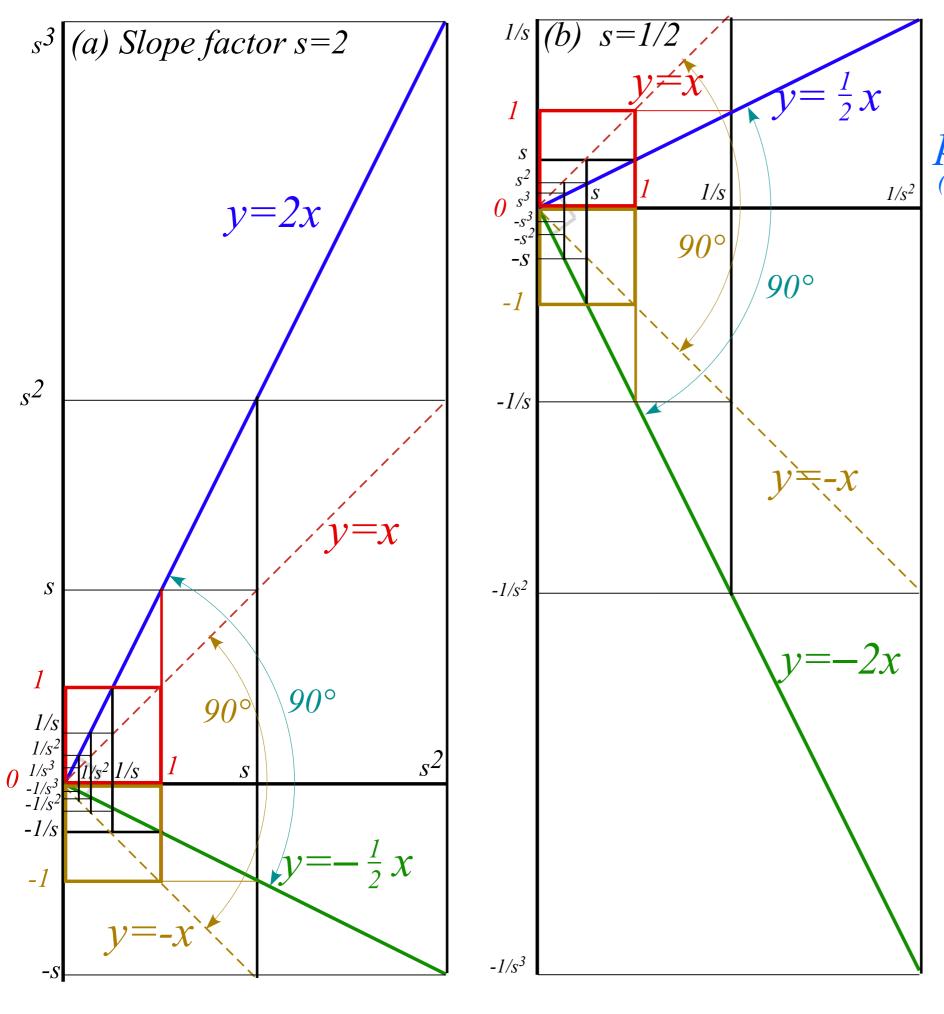
Geometric (Power) series

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Compare mks units of Coulomb Electrostatic vs. Gravity

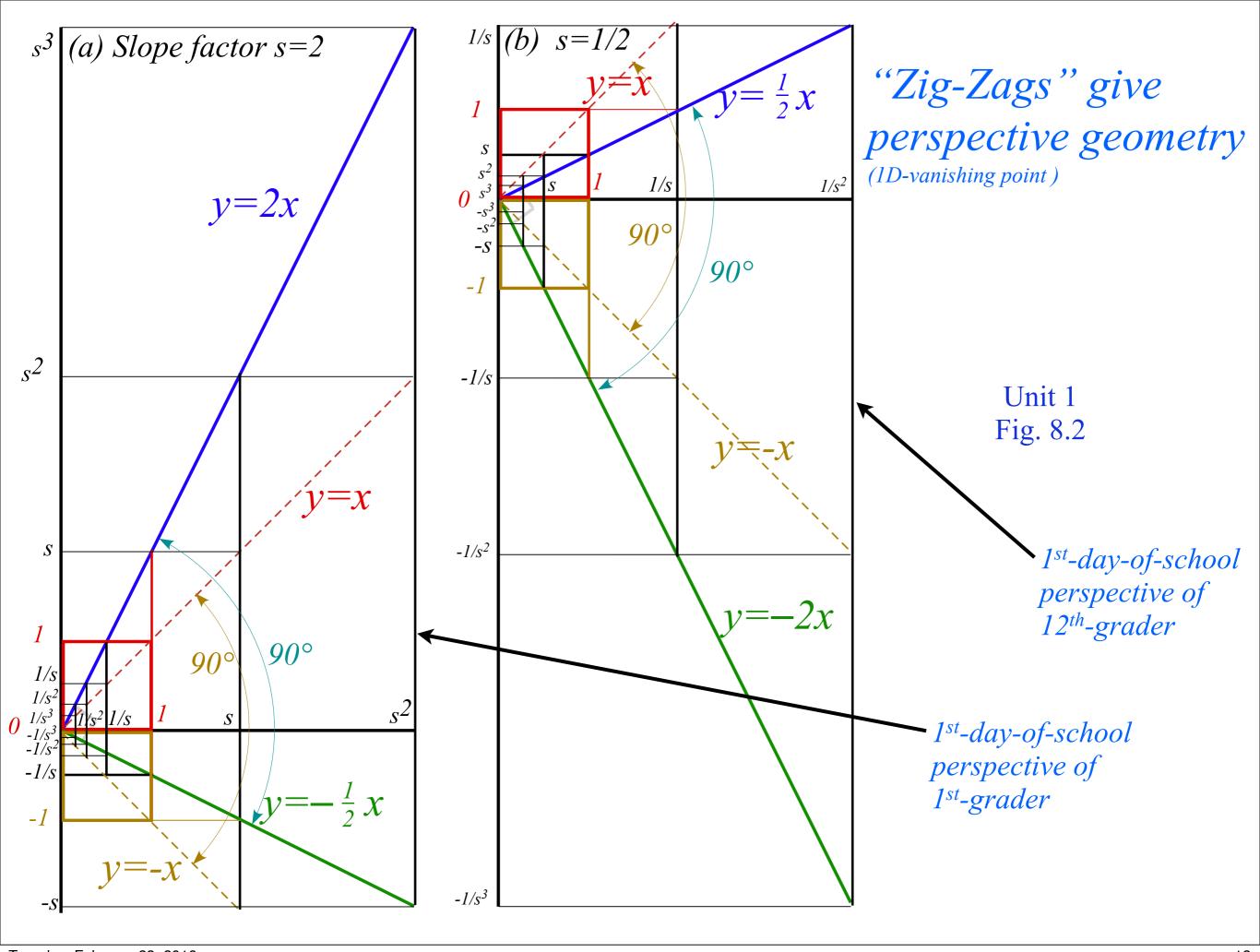


"Zig-Zags" give

perspective geometry

(1D-vanishing point)

Unit 1 Fig. 8.2



Geometry of common power-law potentials

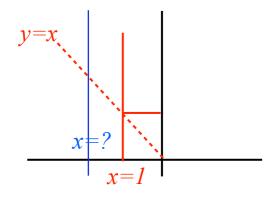
Geometric (Power) series

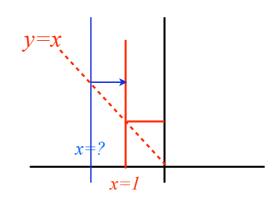
"Zig-Zag" exponential geometry

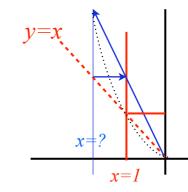
Projective or perspective geometry

Parabolic geometry of harmonic oscillator kr²/2 potential and -kr¹ force fields Coulomb geometry of -1/r-potential and -1/r²-force fields Compare mks units of Coulomb Electrostatic vs. Gravity

- 1. Pick an (x=?)-line 2. "Zig" from its y=xintersection to x=1 line
 - 3. "Zag" from origin back to (x=?)-line

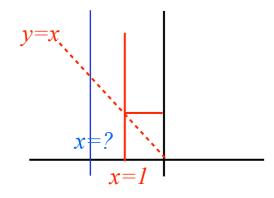


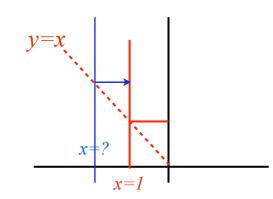


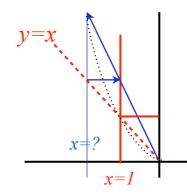


"Zag" line is $y=(?)\cdot x$ and hits (x=?)-line at $y=(?)\cdot(?)=(?)^2$

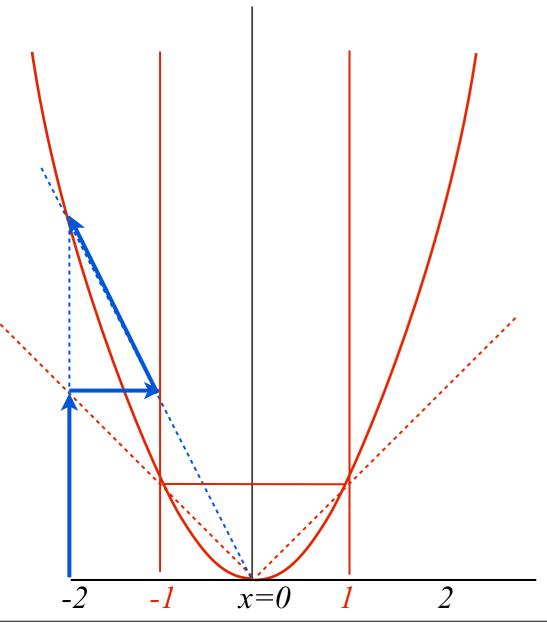
- intersection to x=1 line back to (x=?)-line
- 1. Pick an (x=?)-line 2. "Zig" from its y=x 3. "Zag" from origin





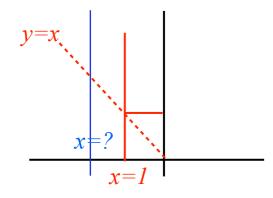


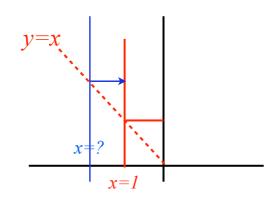
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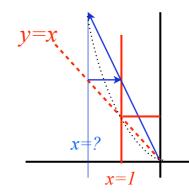


Unit 1 Fig. 9.1

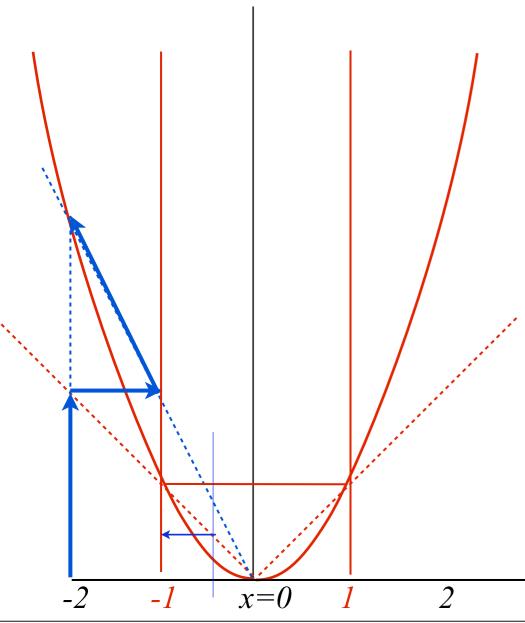
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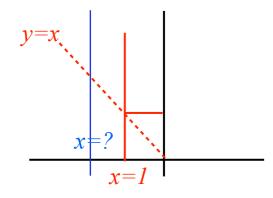


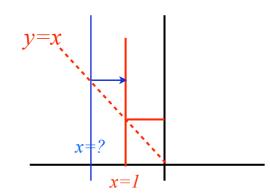
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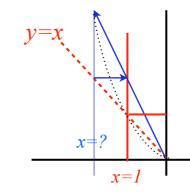


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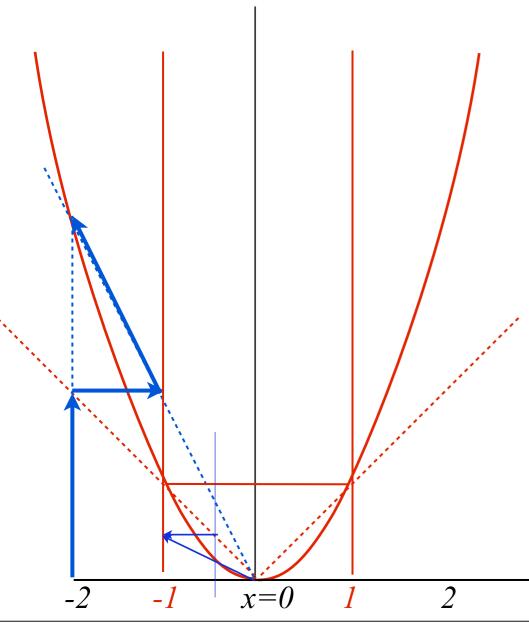
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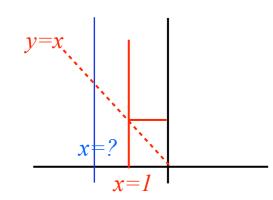


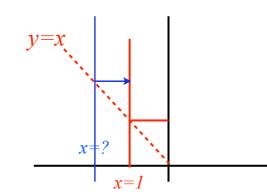
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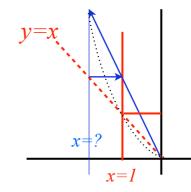


Unit 1 Fig. 9.1

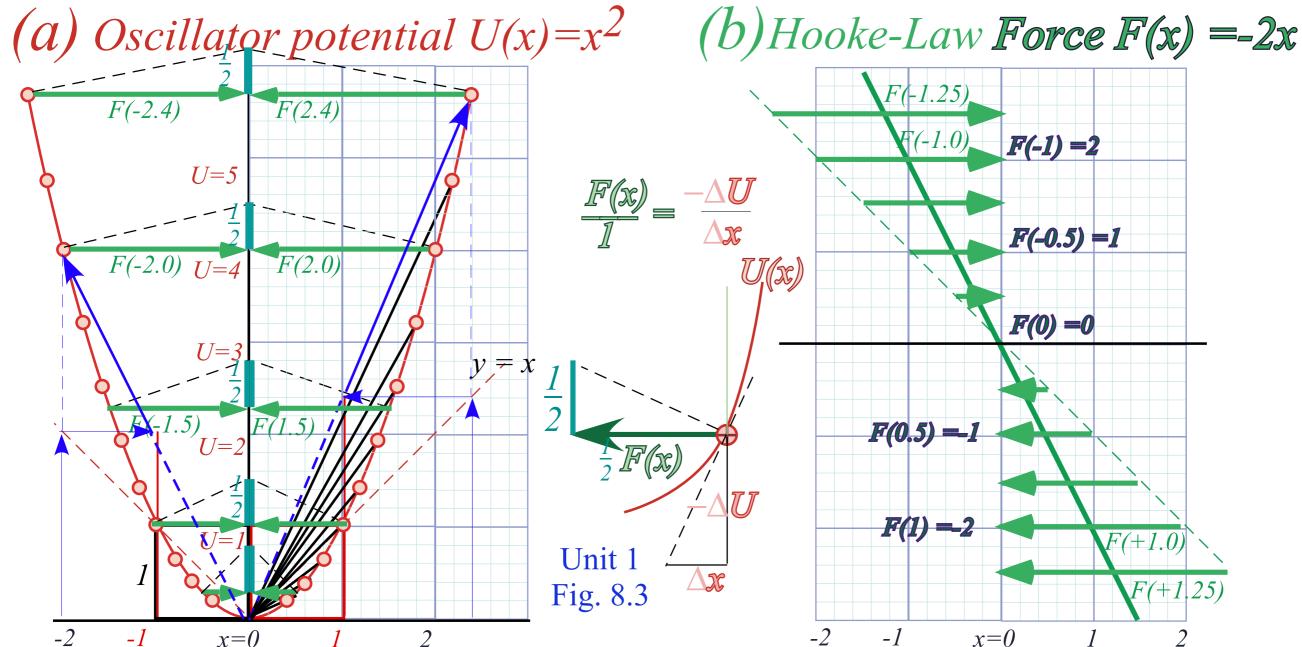
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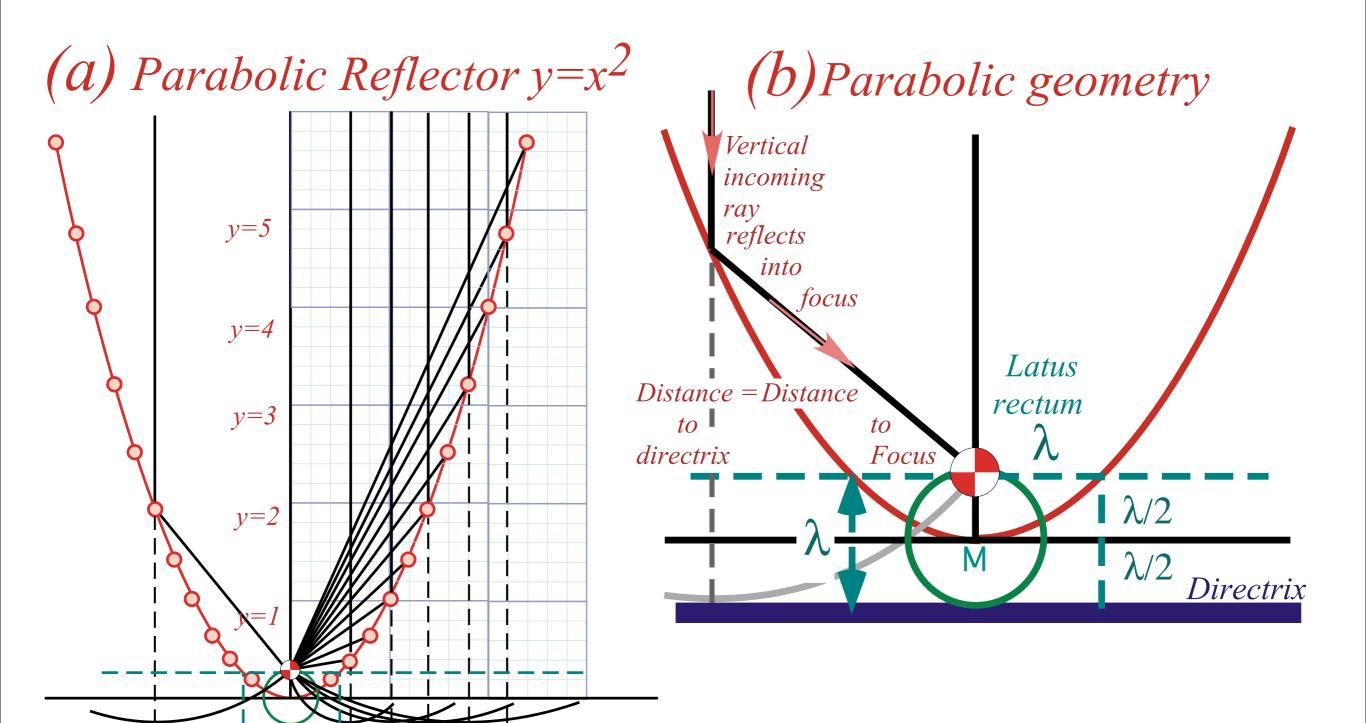




"Zag" line is $y=(?)\cdot x$ and hits (x=?)-line at $y=(?)\cdot(?)=(?)^2$

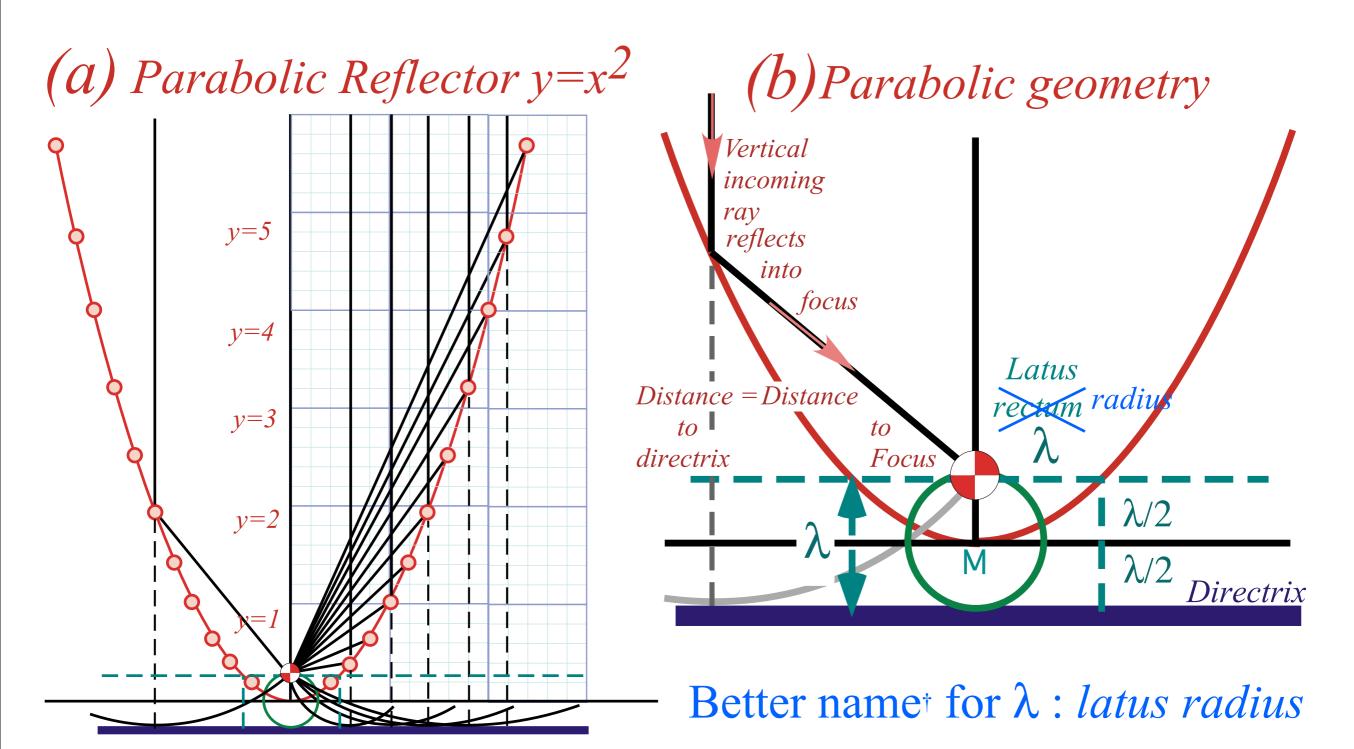


A more conventional parabolic geometry...(uses focal point)



Unit 1 Fig. 8.3

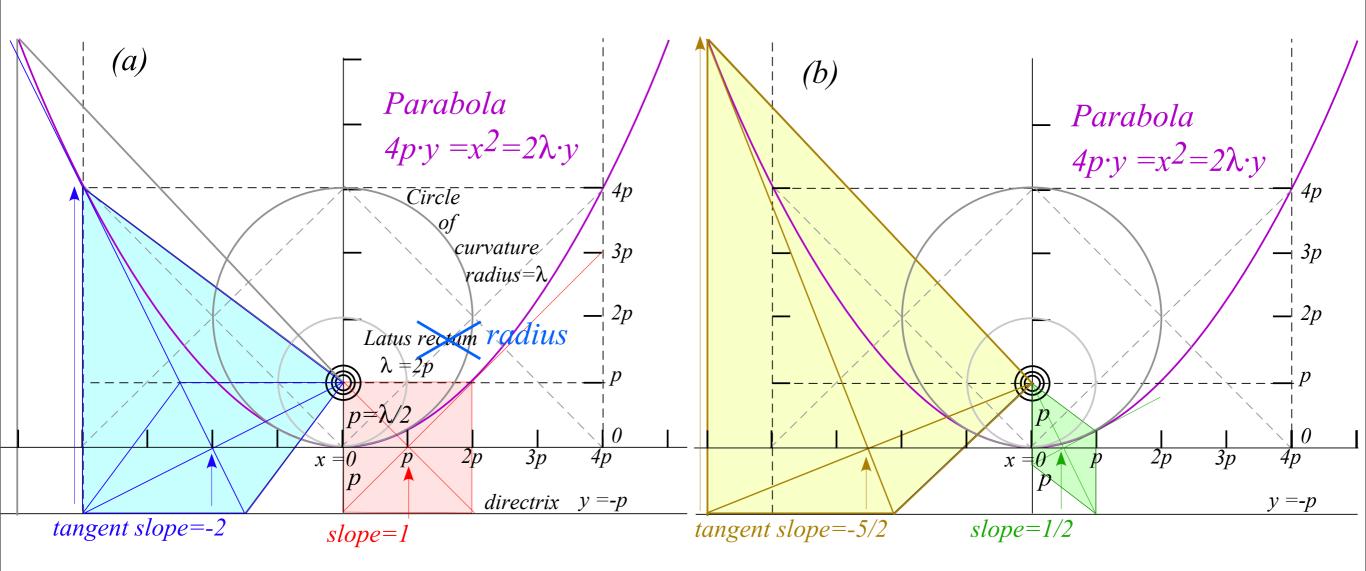
A more conventional parabolic geometry...



Unit 1 Fig. 8.3

Old term *latus rectum* is exclusive copyright of *X-Treme Roidrage Gyms*Venice Beach, CA 90017

...conventional parabolic geometry...carried to extremes...



Unit 1 Fig. 8.4

Geometry of common power-law potentials

Geometric (Power) series

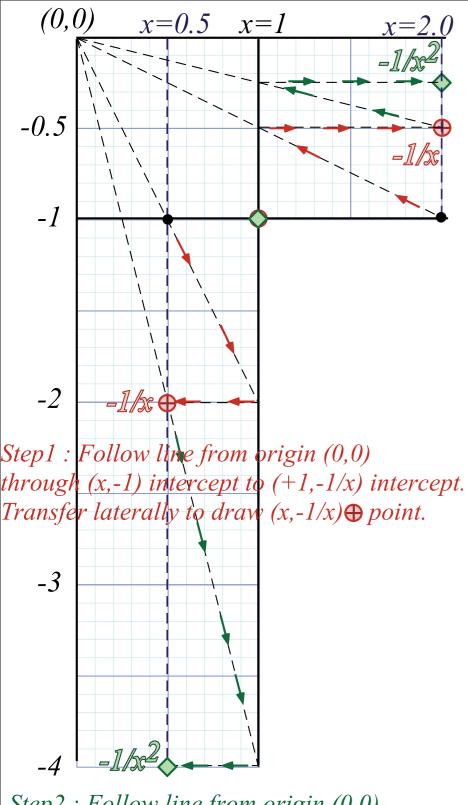
"Zig-Zag" exponential geometry

Projective or perspective geometry

Parabolic geometry of harmonic oscillator kr²/2 potential and -kr¹ force fields

 \rightarrow Coulomb geometry of -1/r-potential and -1/r²-force fields

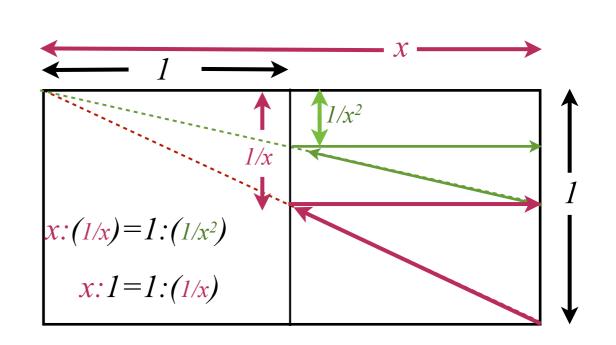
Compare mks units of Coulomb Electrostatic vs. Gravity

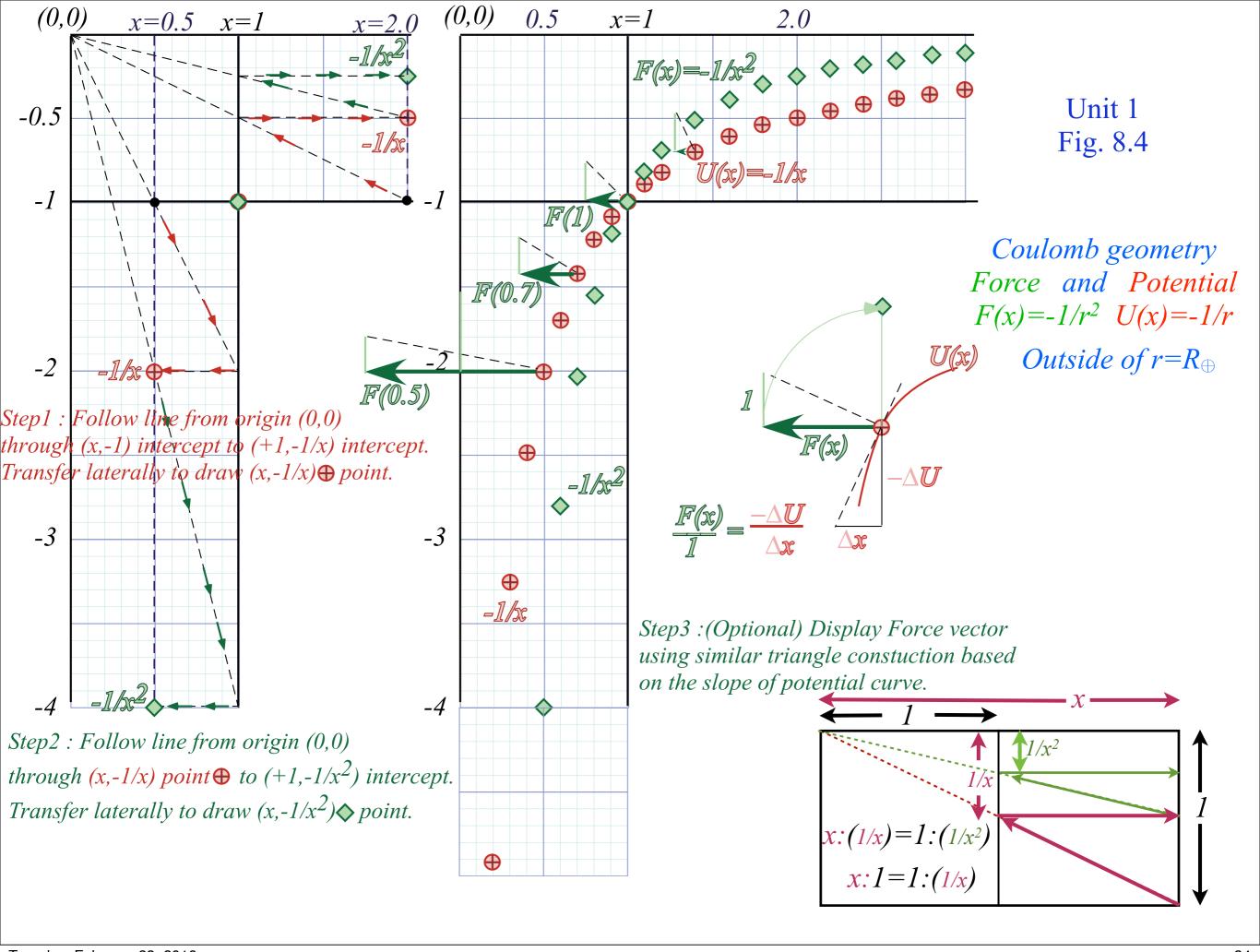


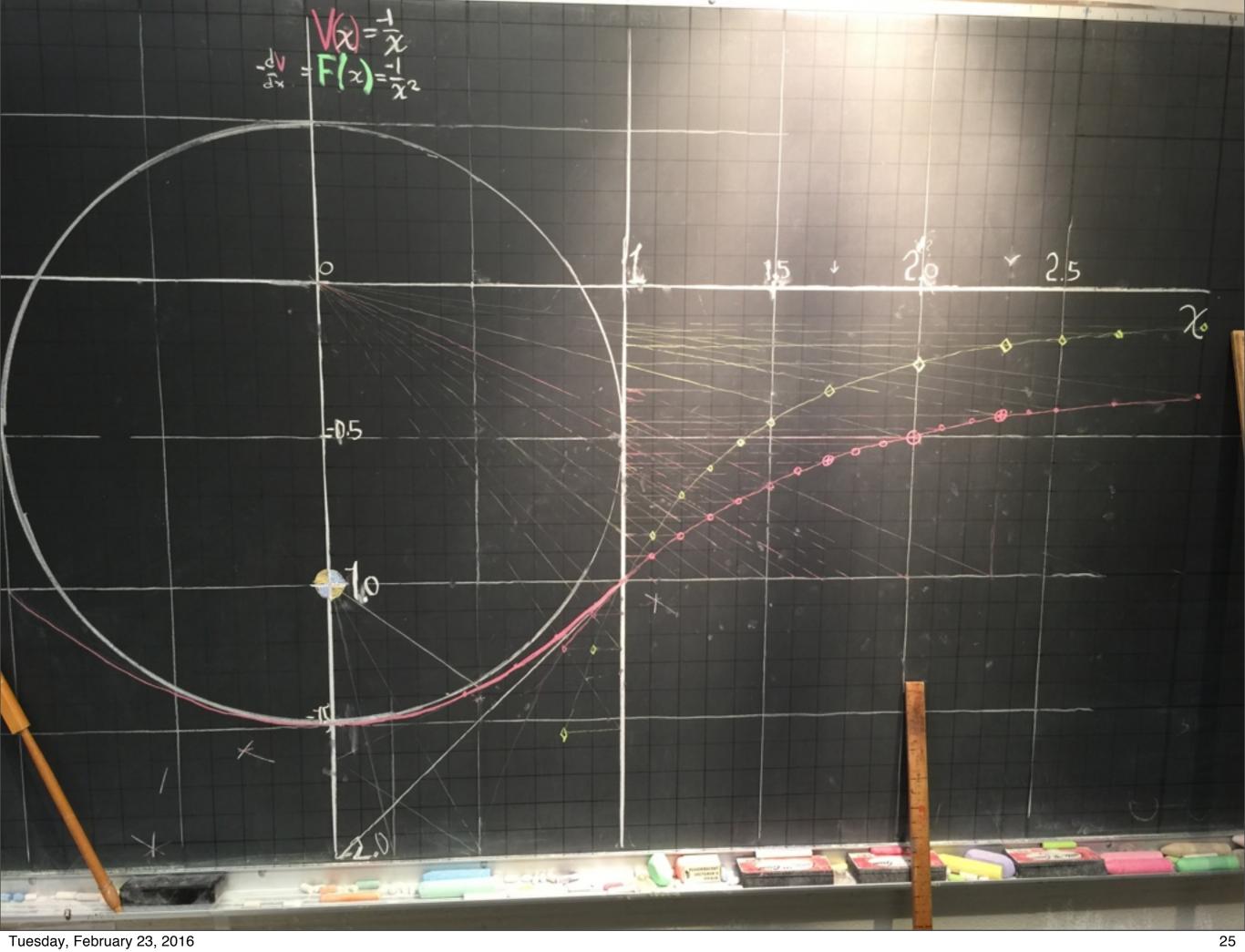
Step2: Follow line from origin (0,0)through (x,-1/x) point \oplus to $(+1,-1/x^2)$ intercept. Transfer laterally to draw $(x,-1/x^2)$ point.

Unit 1 Fig. 8.4

Coulomb geometry
Force and Potential $F(x)=-1/r^2$ U(x)=-1/r







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Parabolic geometry of harmonic oscillator kr²/2 potential and -kr¹ force fields

Coulomb geometry of -1/r-potential and -1/r²-force fields

Compare mks units of Coulomb Electrostatic vs. Gravity

1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong \frac{?.?\cdot 10^?}{per \ square \ Coulomb}$$

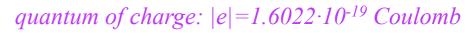
1. Electrostatic force between q(Coulombs) and Q(C.) !!!!

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \approx 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

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More precise value for electrostatic constant: $1/4\pi\epsilon_0=8.987,551\cdot10^9 \text{Nm}^2/\text{C}^2 \sim 9\cdot10^9 \sim 10^{10}$



Repulsive (+)(+) or (-)(-)

Attractive (+)(-) or (-)(+)

...but 1 Ampere = 1 Coulomb/sec.

1. Electrostatic force between q(Coulombs) and Q(C.)

1111

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

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quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb

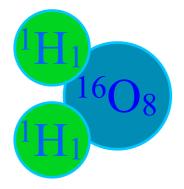
1

Repulsive (+)(+) or (-)(-) Attractive (+)(-) or (-)(+)

 $...but\ 1\ Ampere = 1\ Coulomb/sec.$

"Fingertip Physics" of Ch. 8 notes that $1 \text{ (cm)}^3 = 1\text{gm of water (1/18 Mole) has (1/18)}$ $6 \cdot 10^{23} \text{ molecules}$ $0.3 \cdot 10^{23}$ Number

*H*₂*O* Molecular weight~18



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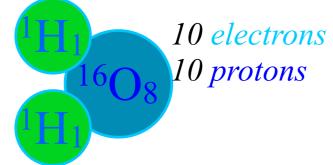
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"Fingertip Physics" of Ch. 9 notes that $1 \text{ (cm)}^3 = 1 \text{gm of water (1/18 Mole) has (1/18) } 6.10^{23} \text{ molecules or } \sim 3.10^{23} \text{ electrons}$ and $\sim 3.10^{23}$ protrons. $\sim 0.3 \cdot 10^{23}$

*H*₂*O Molecular weight~18* Atomic number = 10



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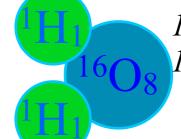
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*H*₂*O* Molecular weight~18 Atomic number = 10



10 electrons That is $\sim -3.10^{23}1.6022\cdot10^{-19}$ Coulomb or about $-0.5\cdot10^{+5}$ C or -50,000 Coulomb plus $\sim +3\cdot 10^{23}1.6022\cdot 10^{-19}$ Coulomb or about $+0.5\cdot 10^{+5}$ C or +50,000 Coulomb 10 protons

Equals zero total charge

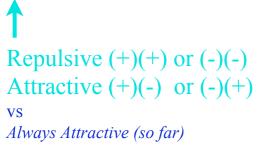
1. Electrostatic force between q(Coulombs) and Q(C.)

////

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

More precise value for electrostatic constant: $1/4\pi\epsilon_0=8.987,551\cdot10^9 \text{Nm}^2/\text{C}^2 \sim 9\cdot10^9 \sim 10^{10}$

quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb







2. Gravitational force between m(kilograms) and M(kg.)

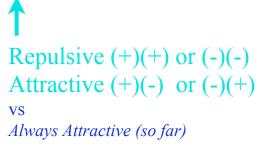
$$F^{grav.}(r) = -G \frac{mM}{r^2} \quad where: G = \underbrace{?.?\cdot 10^?} \qquad \underbrace{Newtons \cdot meter \cdot square}_{per \ square \ kilogram}$$

1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

More precise value for electrostatic constant: $1/4\pi\epsilon_0=8.987,551\cdot10^9 \text{Nm}^2/\text{C}^2 \sim 9\cdot10^9 \sim 10^{10}$

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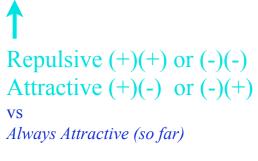
2. Gravitational force between m(kilograms) and M(kg.) !!!!
$$\sim \frac{^{2}/_{3}10^{-10} \sim 10^{-10}}{r^{2}} \text{ where : } G = 0.000,000,000,000,007 \frac{\text{Newtons} \cdot \text{meter} \cdot \text{square}}{\text{per square kilogram}}$$

1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

More precise value for electrostatic constant: $1/4\pi\epsilon_0=8.987,551\cdot10^9 \text{Nm}^2/\text{C}^2 \sim 9\cdot10^9 \sim 10^{10}$

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2. Gravitational force between m(kilograms) and M(kg.)

$$F^{grav.}(r) = -G\frac{mM}{r^2} \quad where: G = 0.000,000,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ kilogram}$$

More precise value for gravitational constant: $G=6.67384(80)\cdot 10^{-11} \text{Nm}^2/\text{C}^2 \sim (2/3)10^{-10}$

1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$

Repulsive (+)(+) or (-)(-)
Attractive (+)(-) or (-)(+)

Discussion of repulsive force and PE in Ch. 8...

quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb

I(a). Electrostatic potential energy between q(Coulombs) and Q(C.)

$$U(r) = \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r} \quad where: \frac{1}{4\pi\varepsilon_0} \approx 9,000,000,000 \frac{Joule}{per square Coulomb}$$

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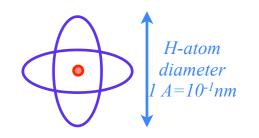
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!!!!!

Nuclear size
$$\sim 10^{-15}$$
 m = 1 femtometer = 1 fm

Atomic size ~ 1 Angstrom = 10^{-10} m



quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb

1. Electrostatic force between q(Coulombs) and Q(C.)

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Repulsive (+)(+) or (-)(-) Attractive (+)(-) or (-)(+) Discussion of repulsive force and PE in Ch. 8...

I(a). Electrostatic potential energy between q(Coulombs) and Q(C.)

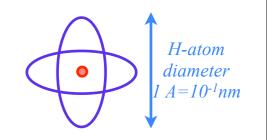
$$U(r) = \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r} \quad where: \frac{1}{4\pi\varepsilon_0} \approx 9,000,000,000 \frac{Joule}{per \ square \ Coulomb}$$
!!!!

Nuclear size $\sim 10^{-15}$ m = 1 femtometer = 1 fm

Atomic size ~ 1 Angstrom = 10^{-10} m

Atomic size ~ 1 Angstrom = 10^{-10} m Big molecule ~ 10 Angstrom = 10^{-9} m = 1nanometer=1nm

quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb



1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$



Repulsive (+)(+) or (-)(-)Attractive (+)(-) or (-)(+)

Discussion of repulsive force and PE in Ch. 8...

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Nuclear size $\sim 10^{-15}$ m = 1 femtometer = 1 fm

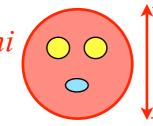
Atomic size ~ 1 Angstrom = 10^{-10} m

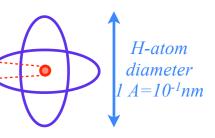
 $Big\ molecule \sim 10\ Angstrom = 10^{-9}\ m = 1nanometer=1nm$

quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb

$$also:1fm = 10^{-13} cm = 1Fermi$$

$$= 1Fm$$





1. Electrostatic force between q(Coulombs) and Q(C.)

$$F^{elec.}(r) = \pm \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r^2} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Newtons \cdot meter \cdot square}{per \ square \ Coulomb}$$



Repulsive (+)(+) or (-)(-)Attractive (+)(-) or (-)(+)

Discussion of repulsive force and PE in Ch. 8...

I(a). Electrostatic potential energy between q(Coulombs) and Q(C.)

$$U(r) = \frac{1}{4\pi\varepsilon_0} \frac{qQ}{r} \quad where: \frac{1}{4\pi\varepsilon_0} \cong 9,000,000,000 \frac{Joule}{per \ square \ Coulomb}$$
!!

Nuclear size $\sim 10^{-15}$ m = 1 femtometer = 1 fm

Atomic size ~ 1 Angstrom = 10^{-10} m

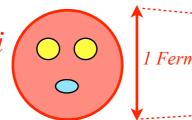
 $Big\ molecule \sim 10\ Angstrom = 10^{-9}\ m = 1nanometer=1nm$

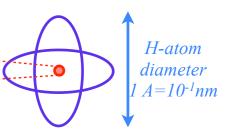
quantum of charge: $|e|=1.6022\cdot 10^{-19}$ Coulomb

$$also:1fm = 10^{-13} cm = 1Fermi$$

$$=1Fm$$

$$1 Fermi$$





nuclear radii are 100,000 to 1,000,000 times smaller than atomic/chemical radii

...so nuclear qQ/r energy 100,000 to 1,000,000 times bigger that of atomic/chemical...

Geometry of idealized "Sophomore-physics Earth"

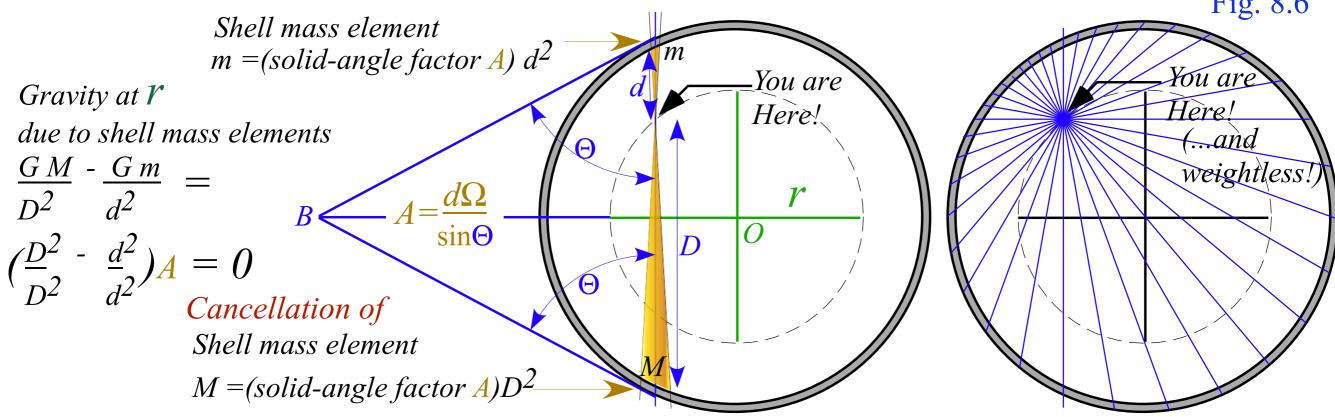
Coulomb field <u>outside</u> Isotropic Harmonic Oscillator (IHO) field <u>inside</u>
Contact-geometry of potential curve(s)

"Crushed-Earth" models: 3 key energy "steps" and 4 key energy "levels" Earth matter vs nuclear matter:

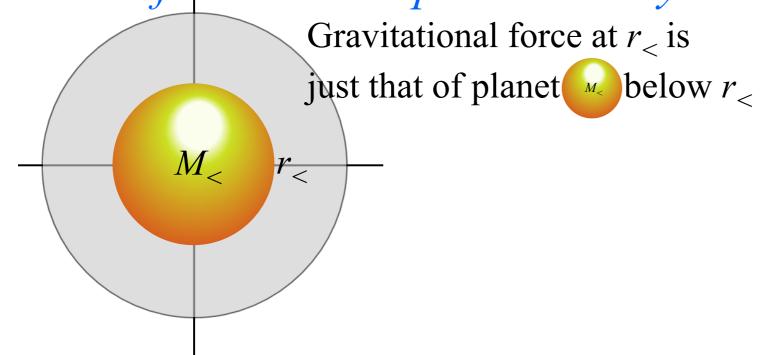
Introducing the "neutron starlet" and "Black-Hole-Earth"

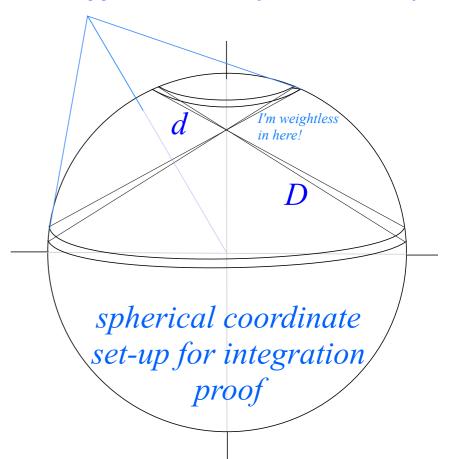


Unit 1 Fig. 8.6



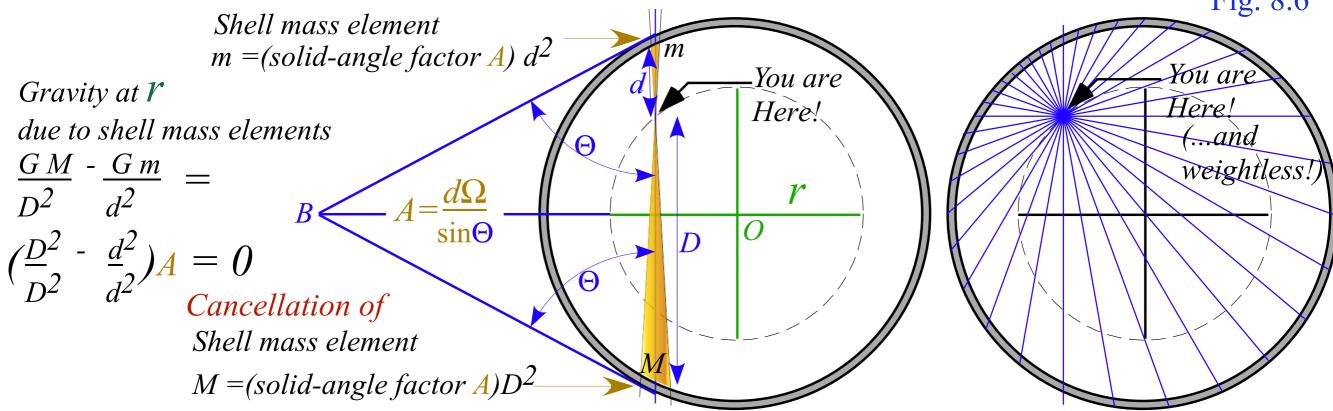
Coulomb force inside-spherical body due to stuff below you, only.



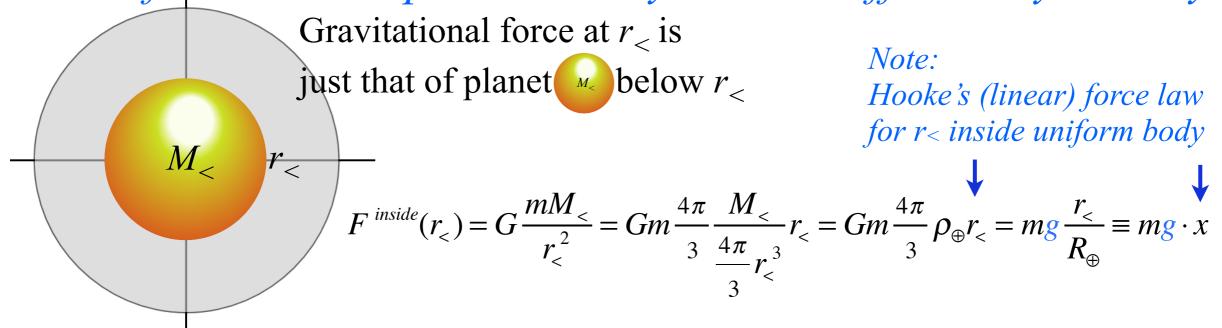




Unit 1 Fig. 8.6

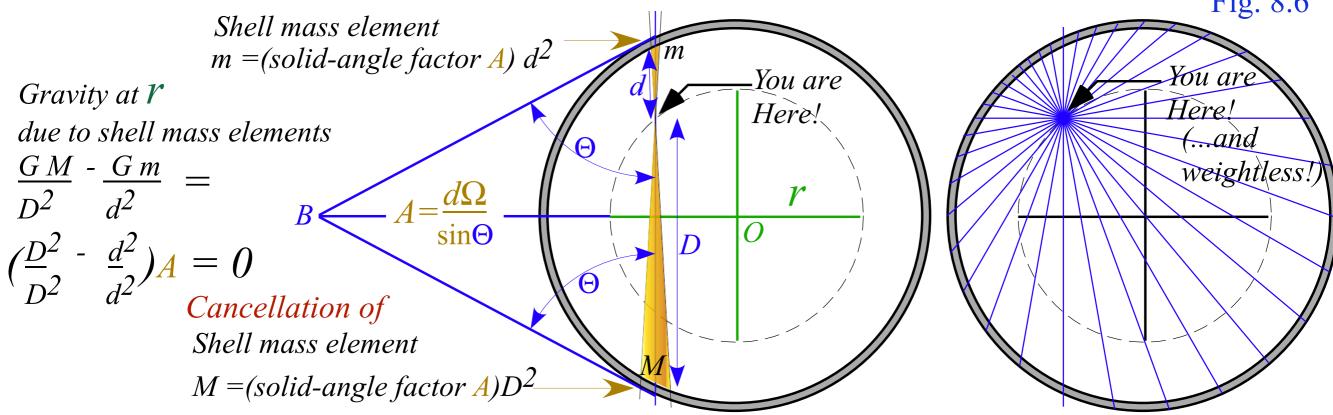


Coulomb force inside-spherical body due to stuff below you, only.

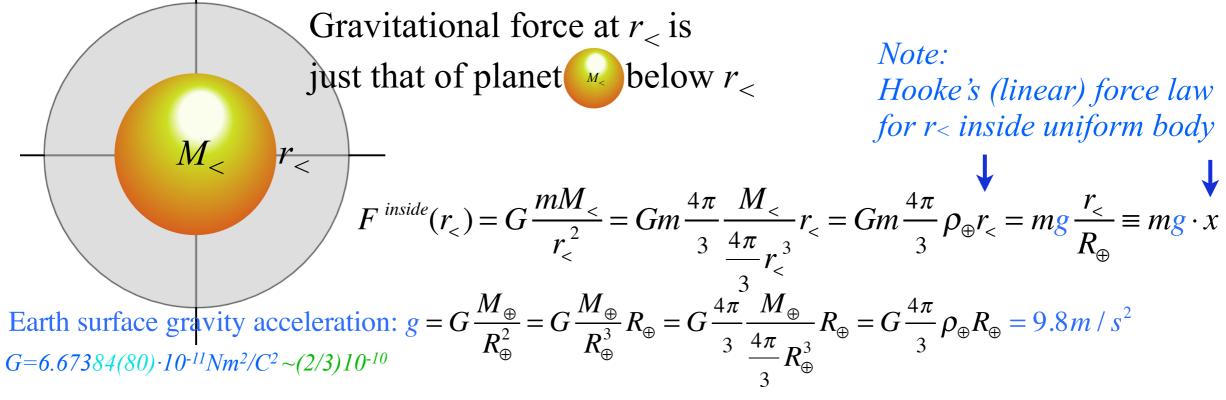




Unit 1 Fig. 8.6

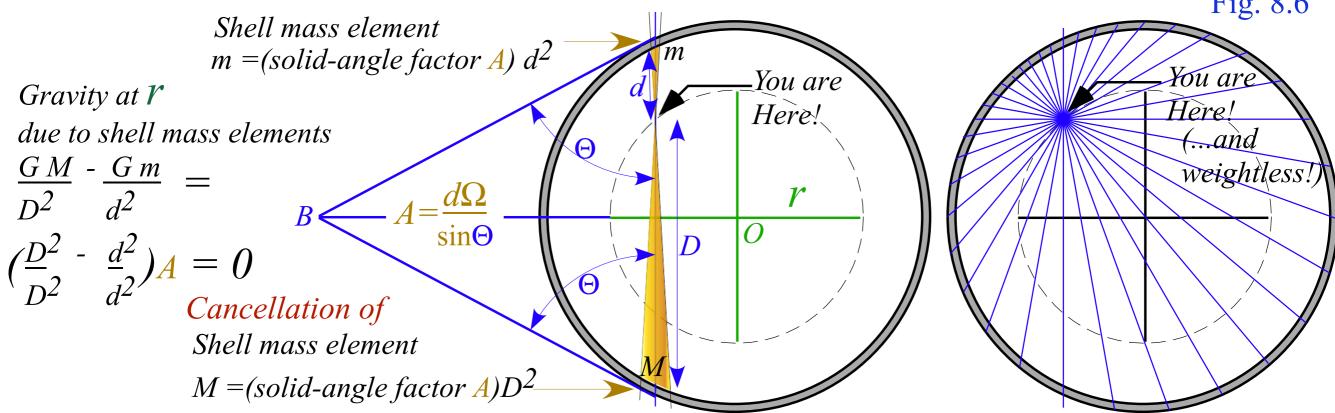


Coulomb force inside-spherical body due to stuff below you, only.

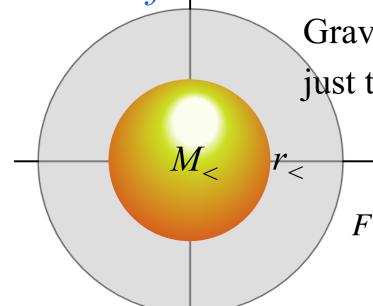


Coulomb force vanishes inside-spherical shell (Gauss-law)

Unit 1 Fig. 8.6



Coulomb force inside-spherical body due to stuff below you, only.



Gravitational force at $r_{<}$ is just that of planet below $r_{<}$

Note:

Hooke's (linear) force law for r< inside uniform body

$$F^{inside}(r_{<}) = G \frac{mM_{<}}{r_{<}^{2}} = Gm \frac{4\pi}{3} \frac{M_{<}}{\frac{4\pi}{2} r_{<}^{3}} r_{<} = Gm \frac{4\pi}{3} \rho_{\oplus} r_{<} = mg \frac{r_{<}}{R_{\oplus}} \equiv mg \cdot x$$

Earth surface gravity acceleration:
$$g = G \frac{M_{\oplus}}{R_{\oplus}^2} = G \frac{M_{\oplus}}{R_{\oplus}^3} R_{\oplus} = G \frac{4\pi}{3} \frac{M_{\oplus}}{4\pi} R_{\oplus} = G \frac{4\pi}{3} \rho_{\oplus} R_{\oplus} = 9.8 m / s^2$$

$$G = 6.67384(80) \cdot 10^{-11} \text{Nm}^2/\text{C}^2 \sim (2/3)10^{-10}$$

Earth radius: $R_{\oplus} = 6.371 \cdot 10^6 m \approx 6.4 \cdot 10^6 m$

Earth mass: $M_{\oplus} = 5.9722 \times 10^{24} \, kg. \approx 6.0 \cdot 10^{24} \, kg.$

Solar radius: $R_{\odot} = 6.955 \times 10^8 m. \approx 7.0 \cdot 10^8 m.$

Solar mass: $M_{\odot} = 1.9889 \times 10^{30} kg. \approx 2.0 \cdot 10^{30} kg.$

Geometry of idealized "Sophomore-physics Earth"

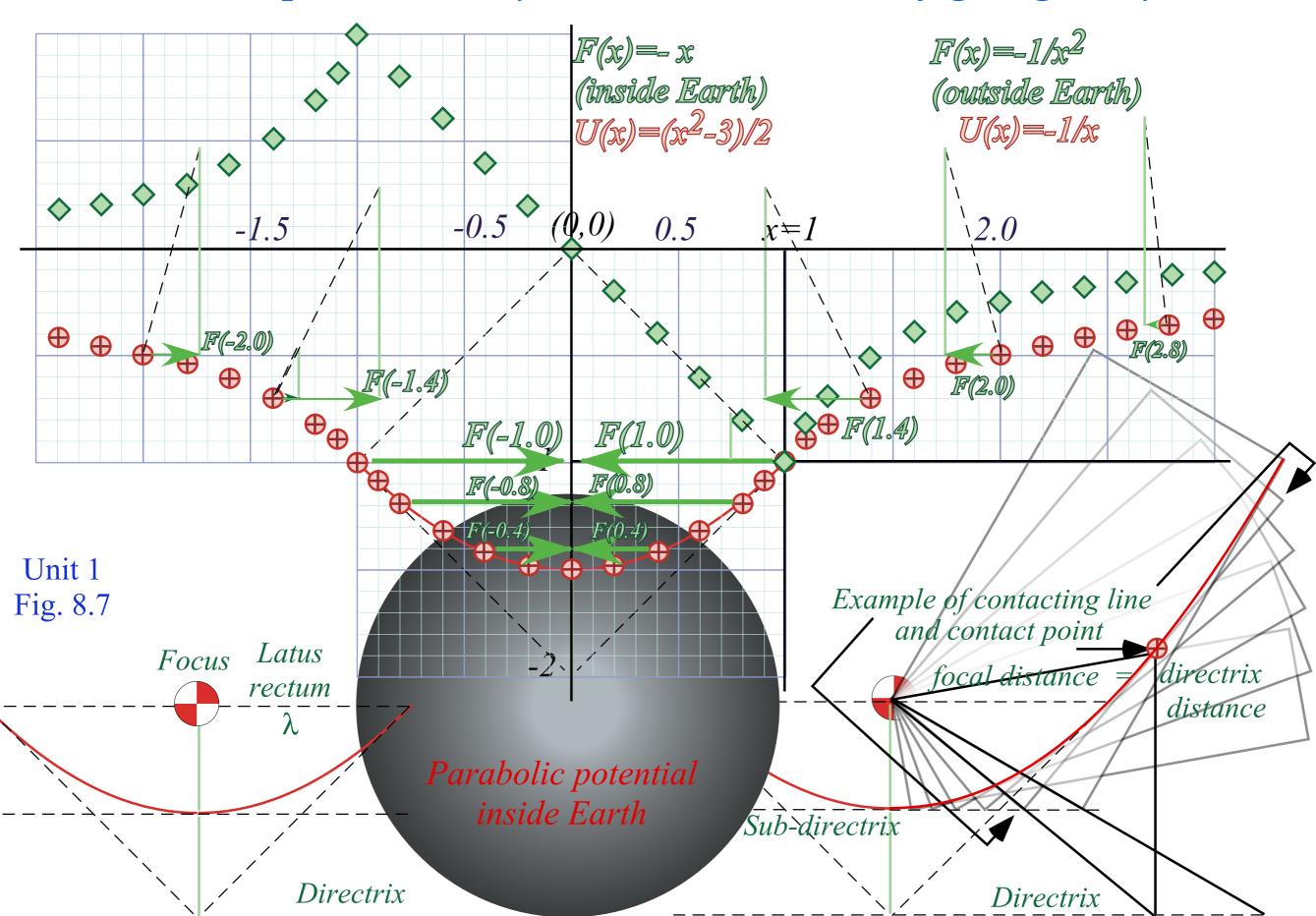
Coulomb field <u>outside</u> Isotropic Harmonic Oscillator (IHO) field <u>inside</u>

Contact-geometry of potential curve(s)

"Crushed-Earth" models: 3 key energy "steps" and 4 key energy "levels" Earth matter vs nuclear matter:

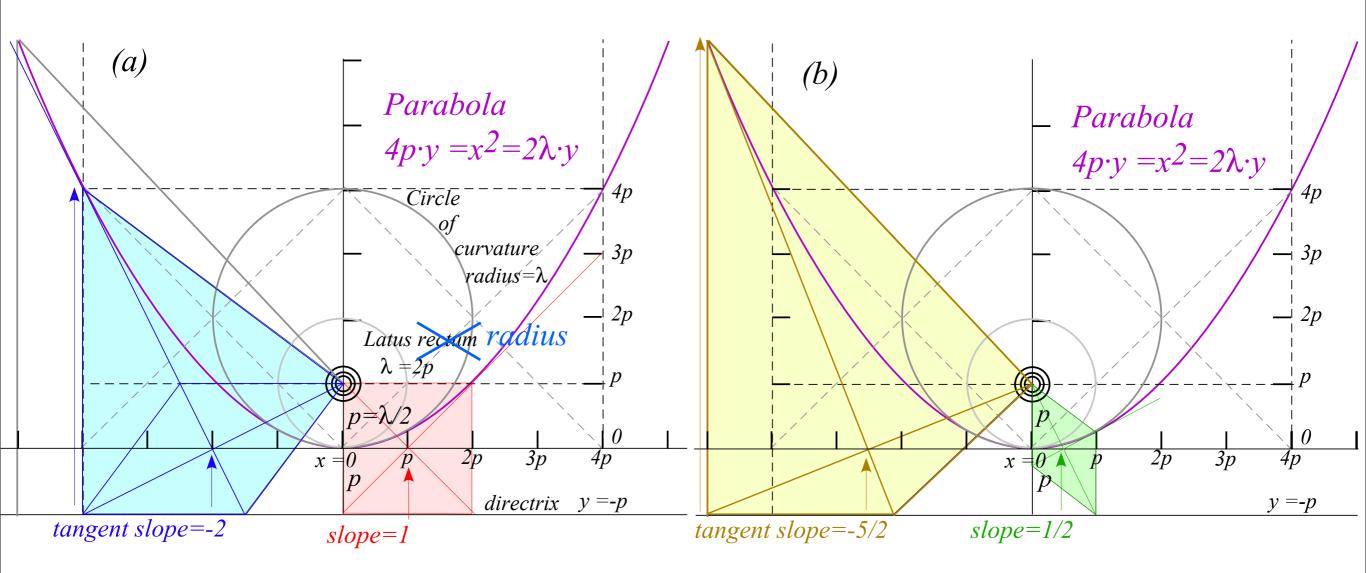
Introducing the "neutron starlet" and "Black-Hole-Earth"

The ideal "Sophomore-Physics-Earth" model of geo-gravity



...conventional parabolic geometry...carried to extremes...

(*From p.21*)



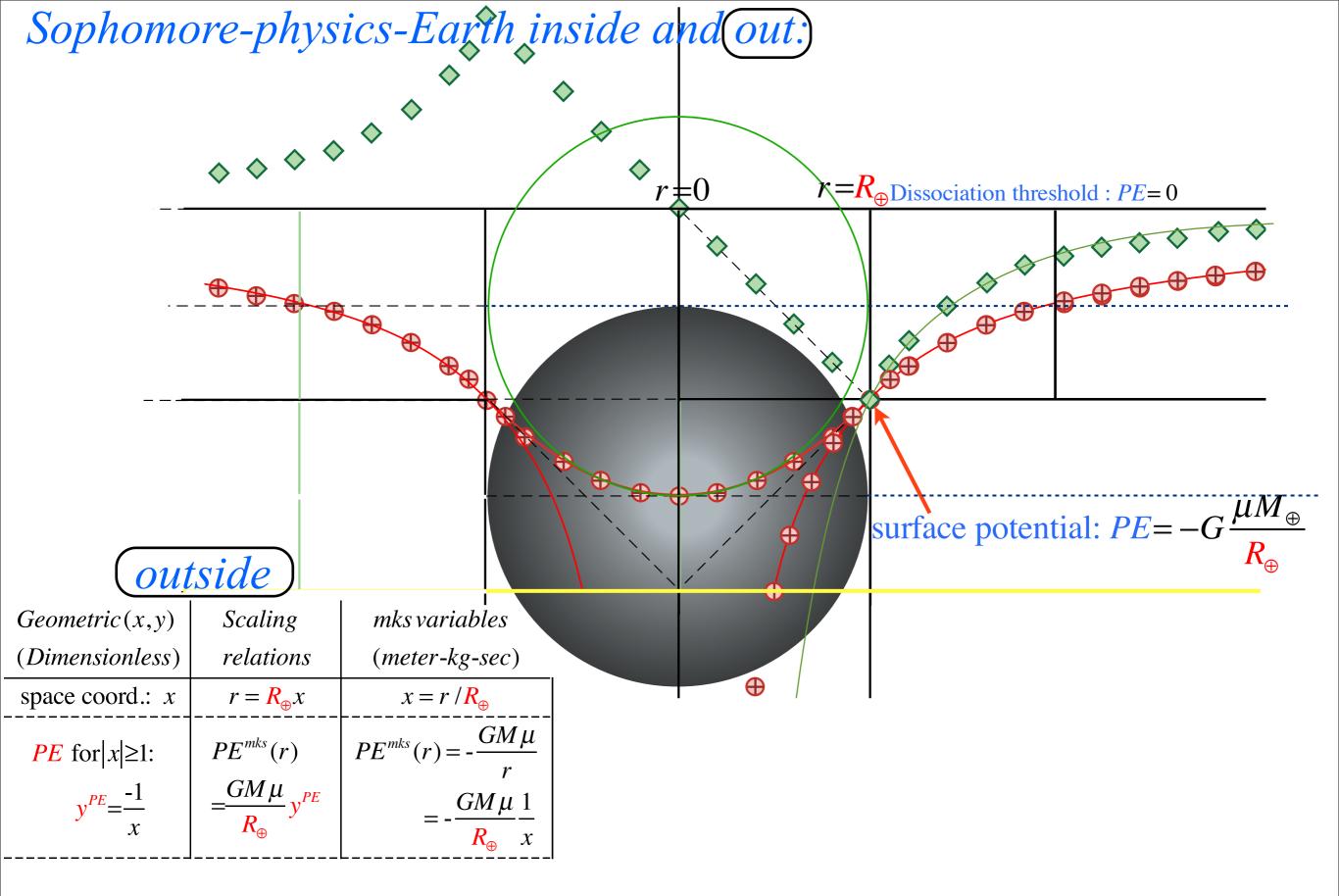
Unit 1 Fig. 8.4

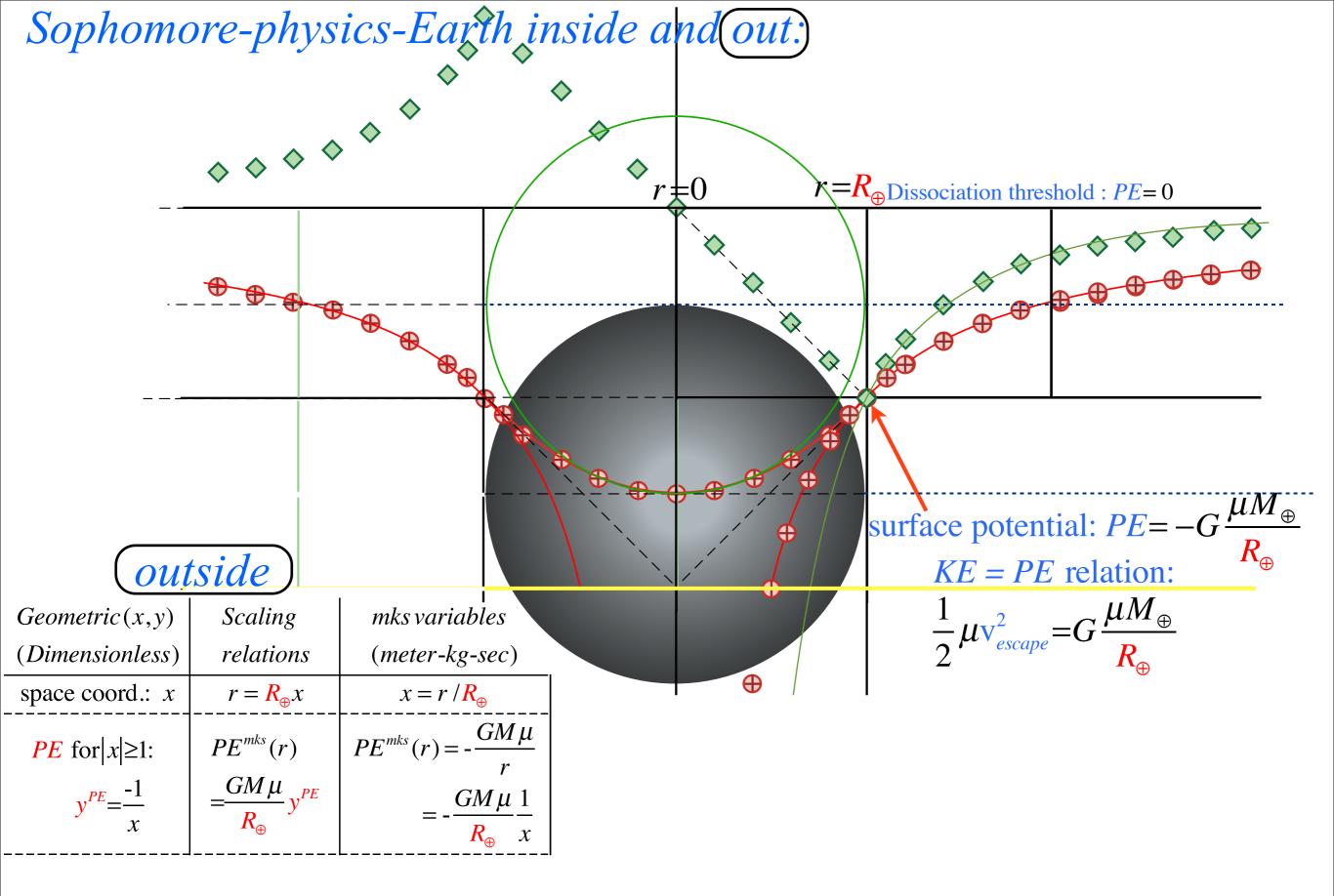
Geometry of idealized "Sophomore-physics Earth"

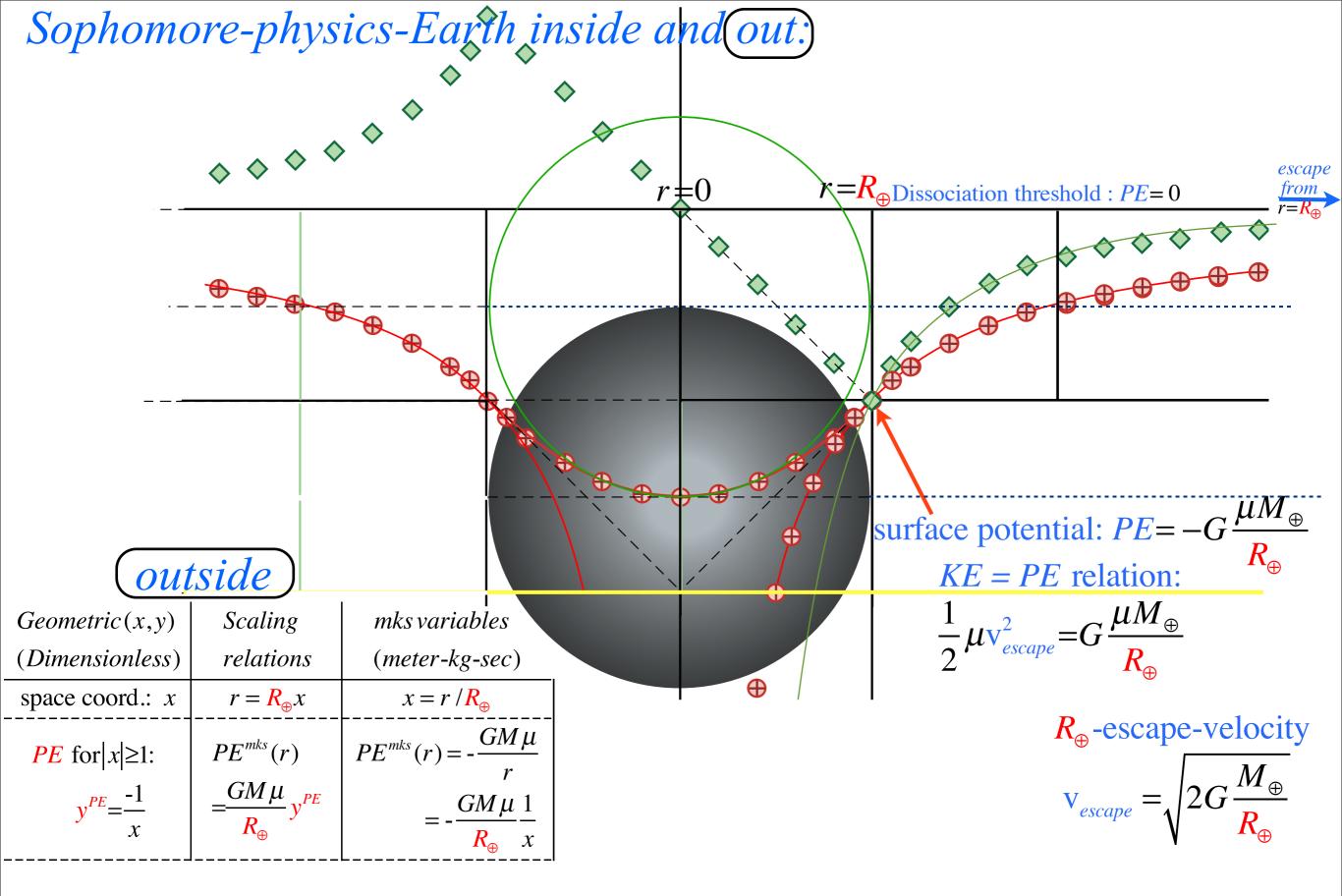
Coulomb field <u>outside</u> Isotropic Harmonic Oscillator (IHO) field <u>inside</u> Contact-geometry of potential curve(s)

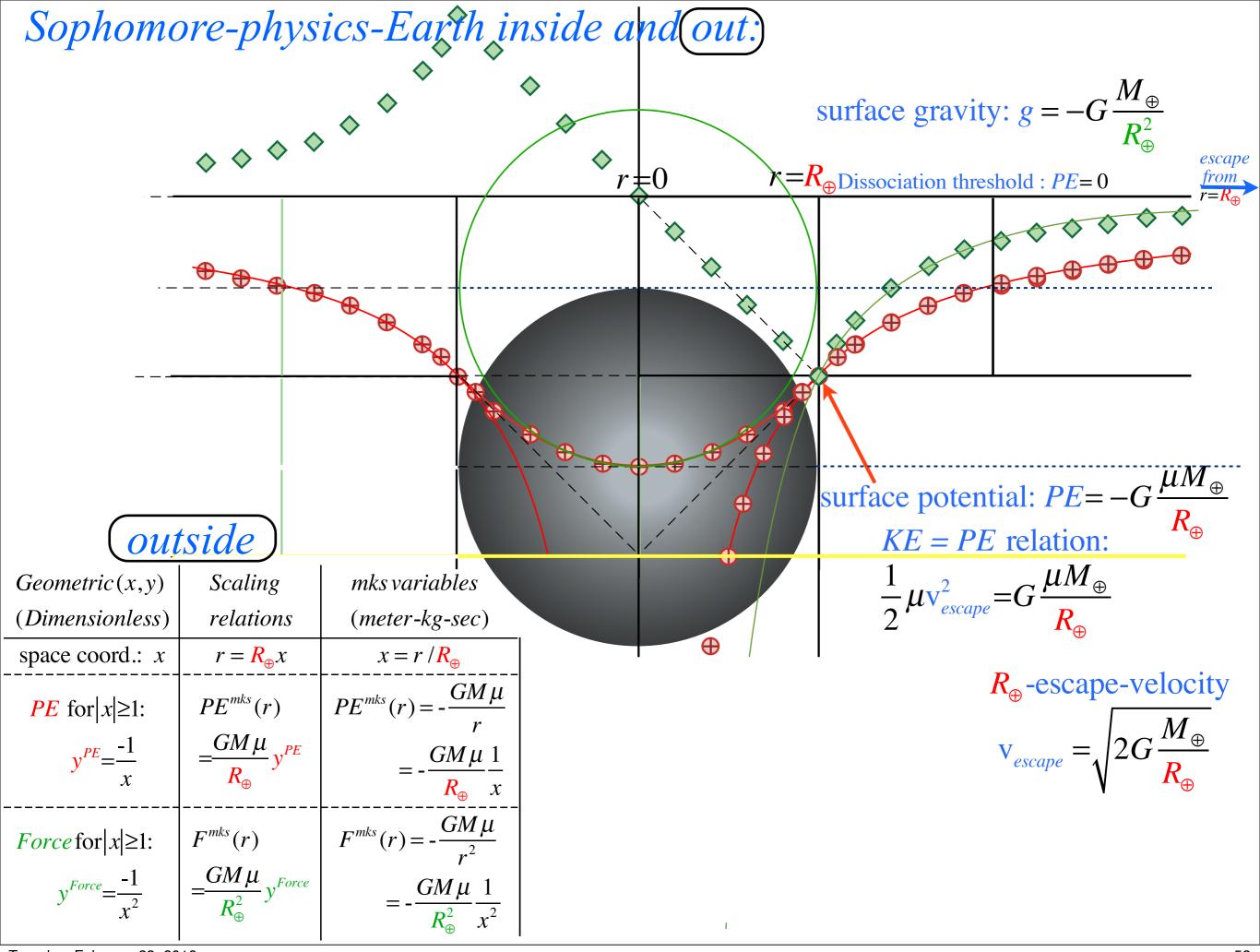
"Crushed-Earth" models: 3 key energy "steps" and 4 key energy "levels" Earth matter vs nuclear matter:

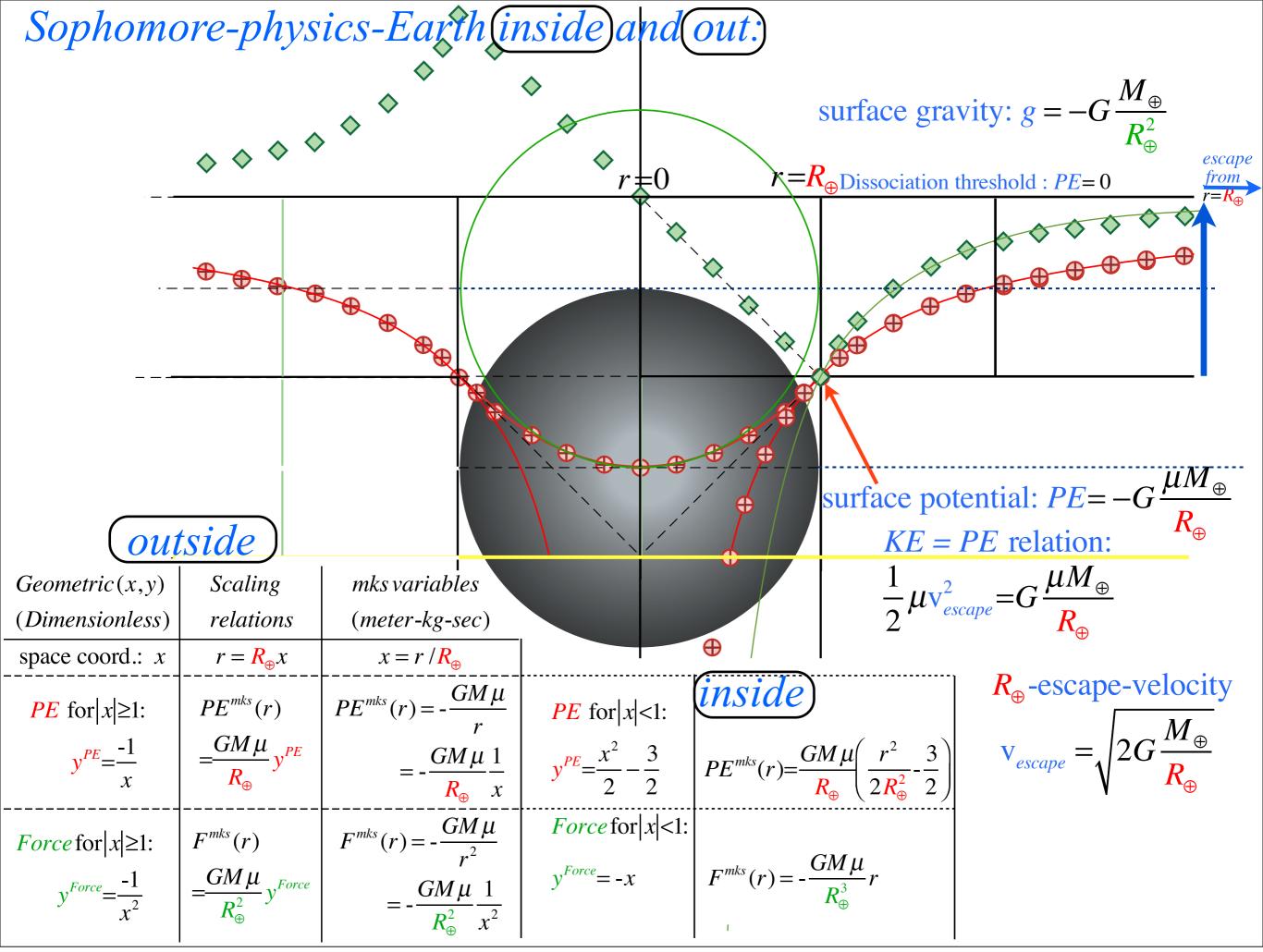
Introducing the "neutron starlet" and "Black-Hole-Earth"

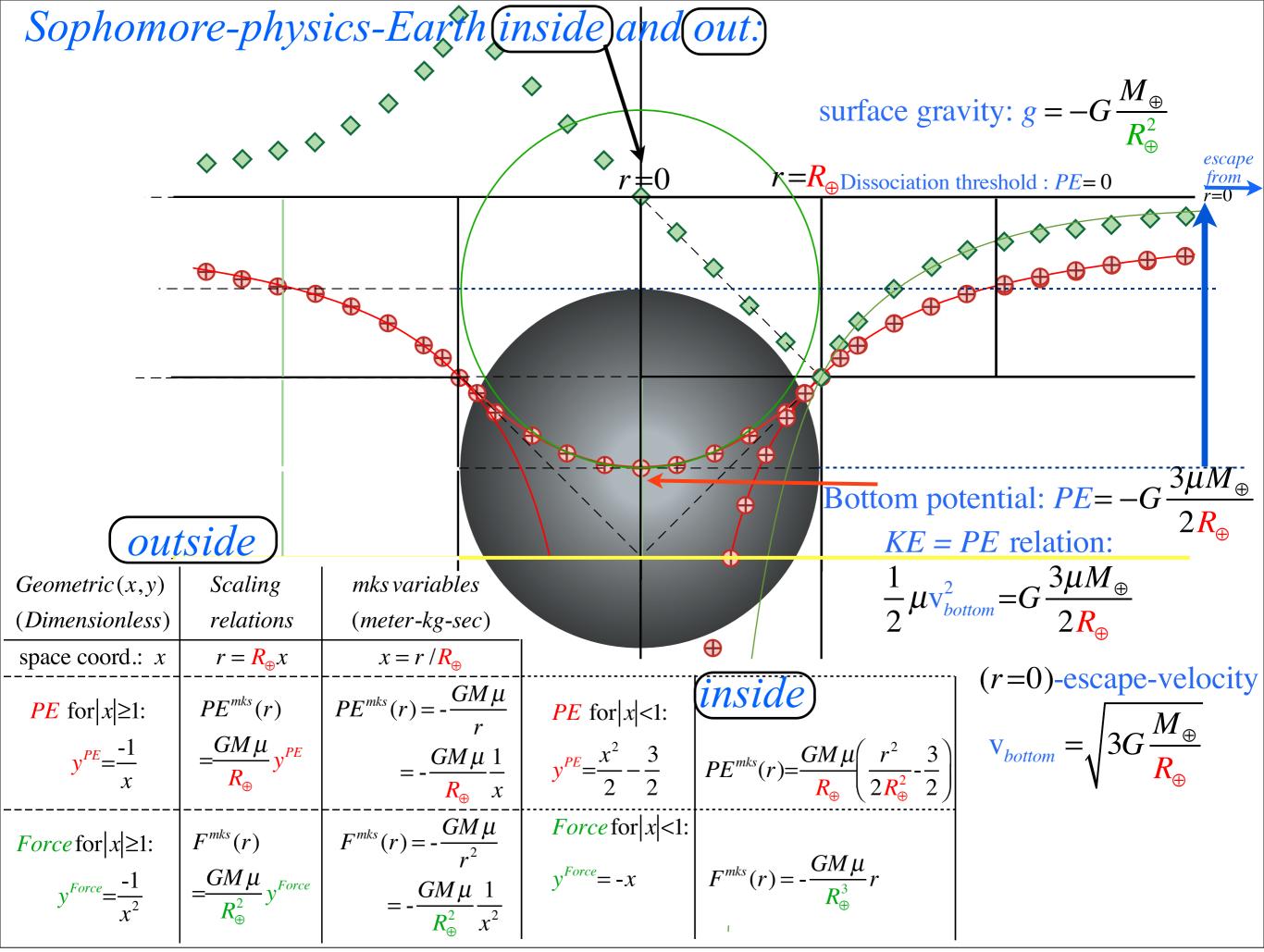


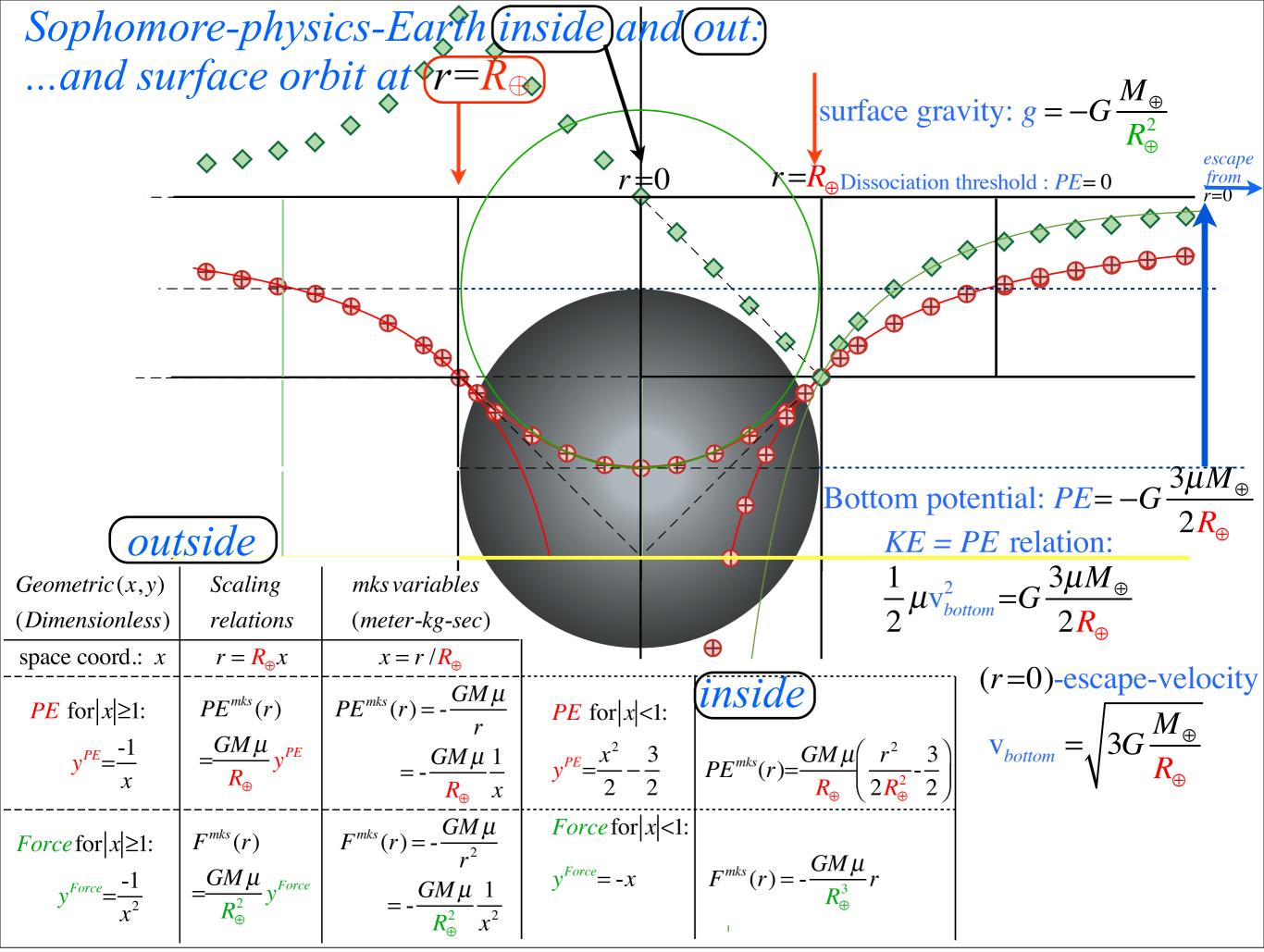


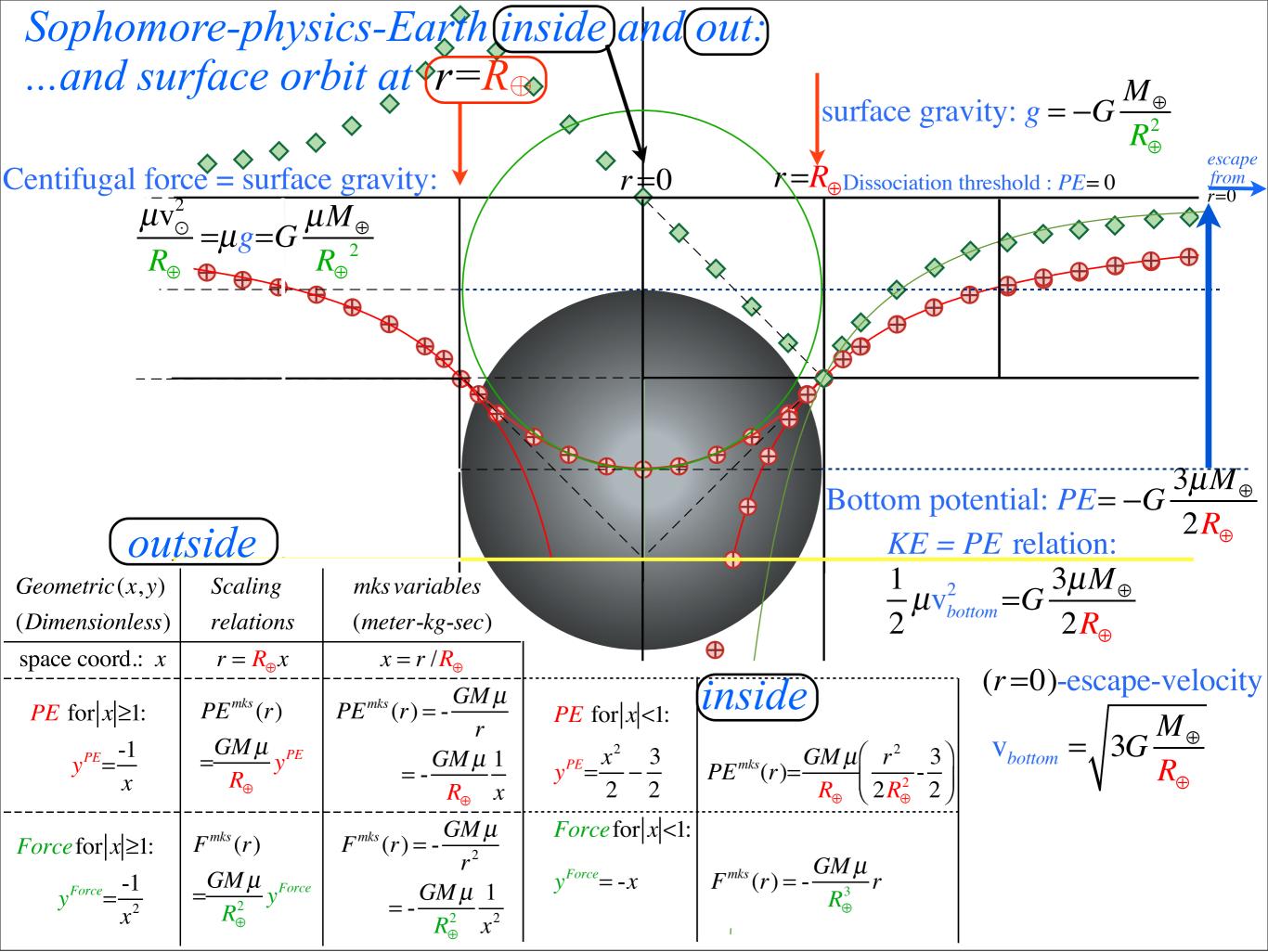


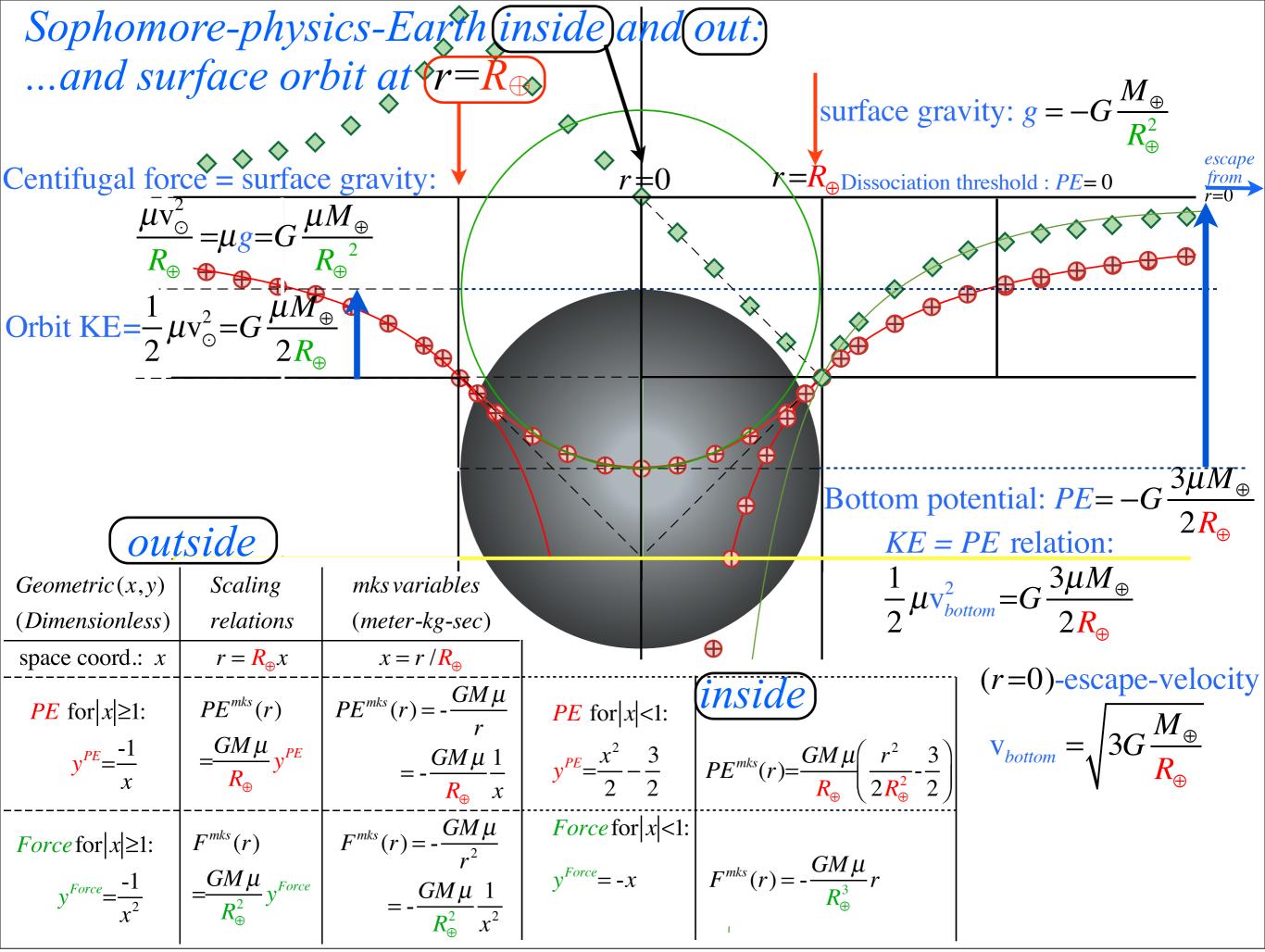


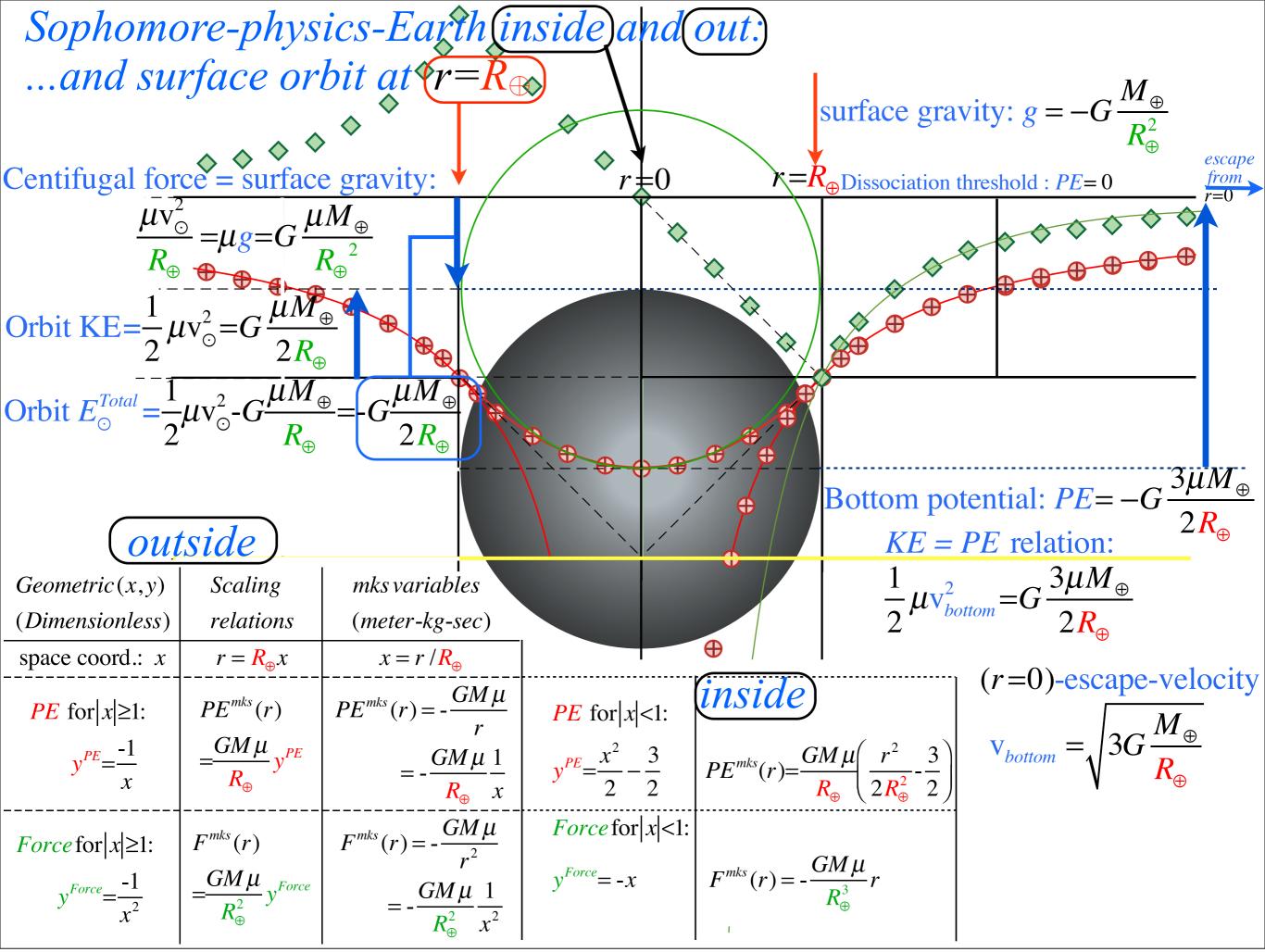


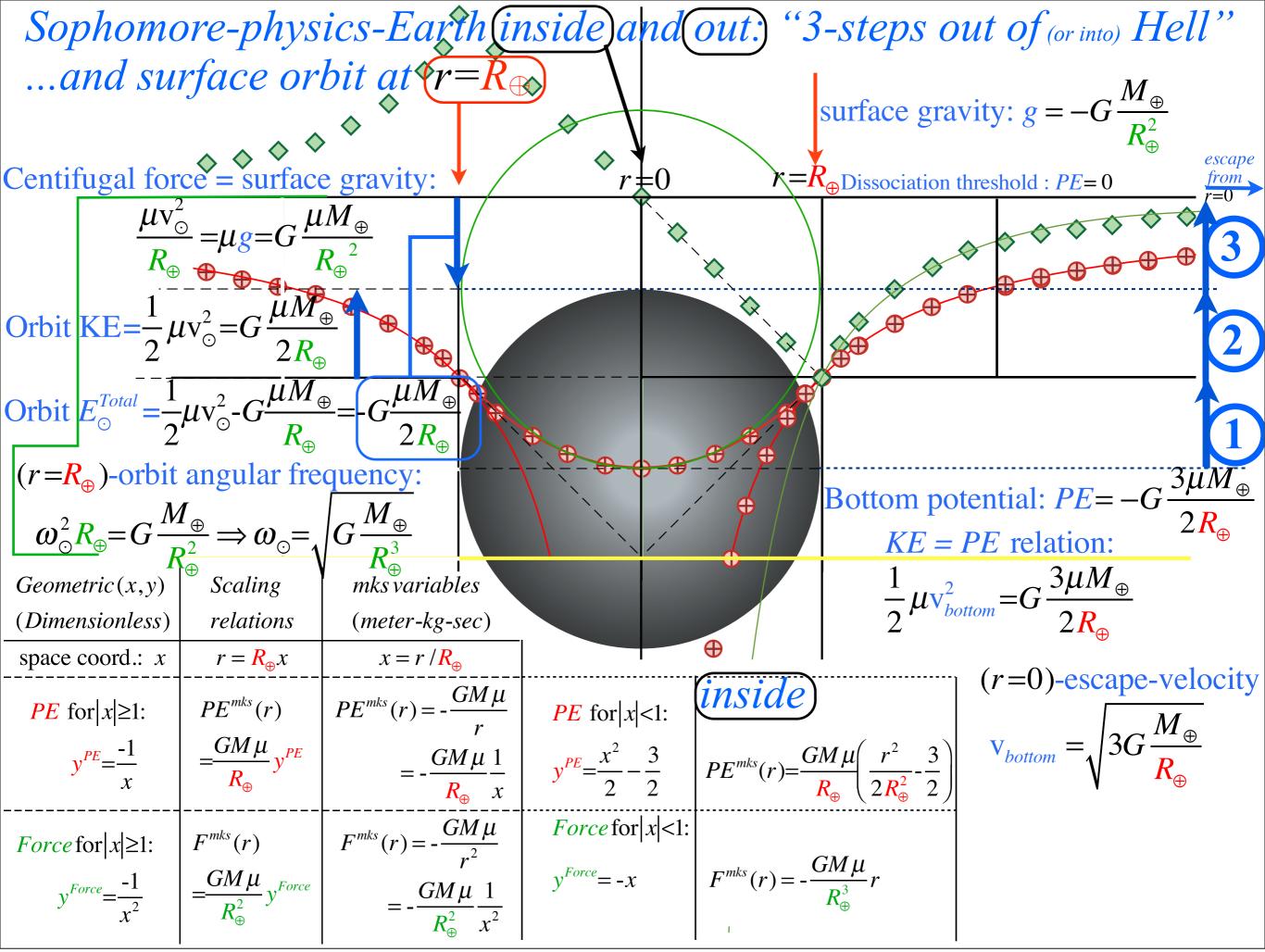


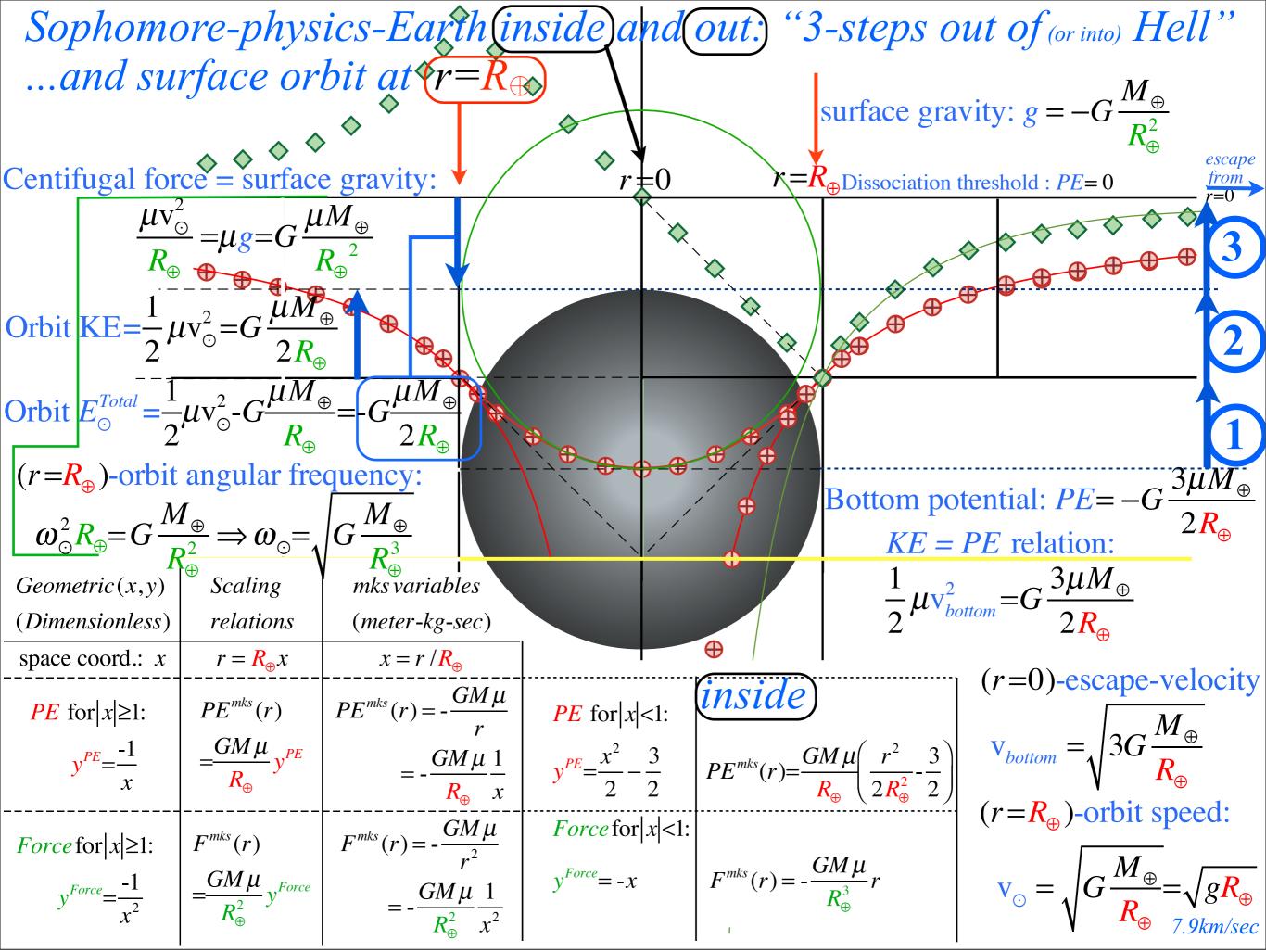


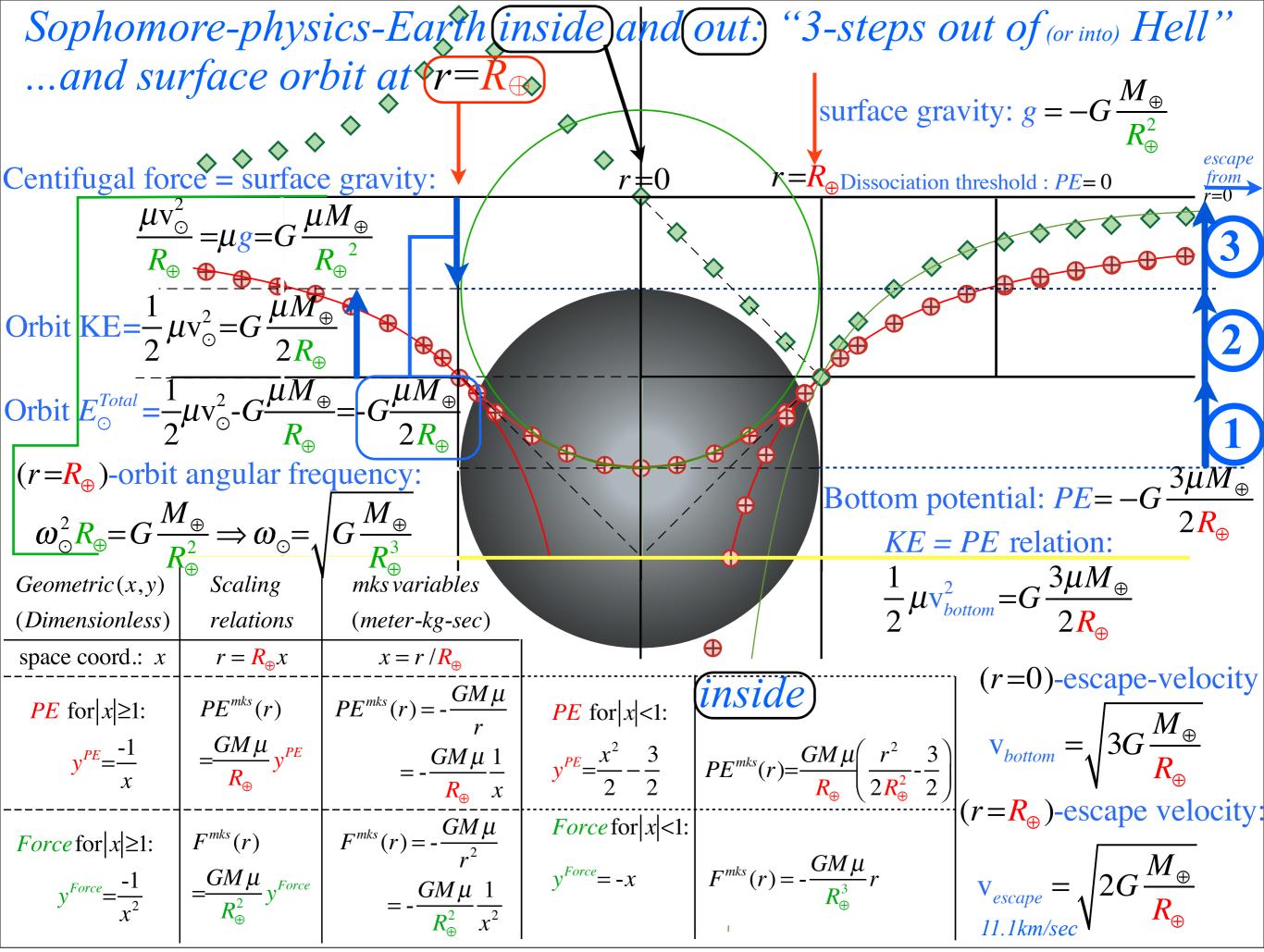


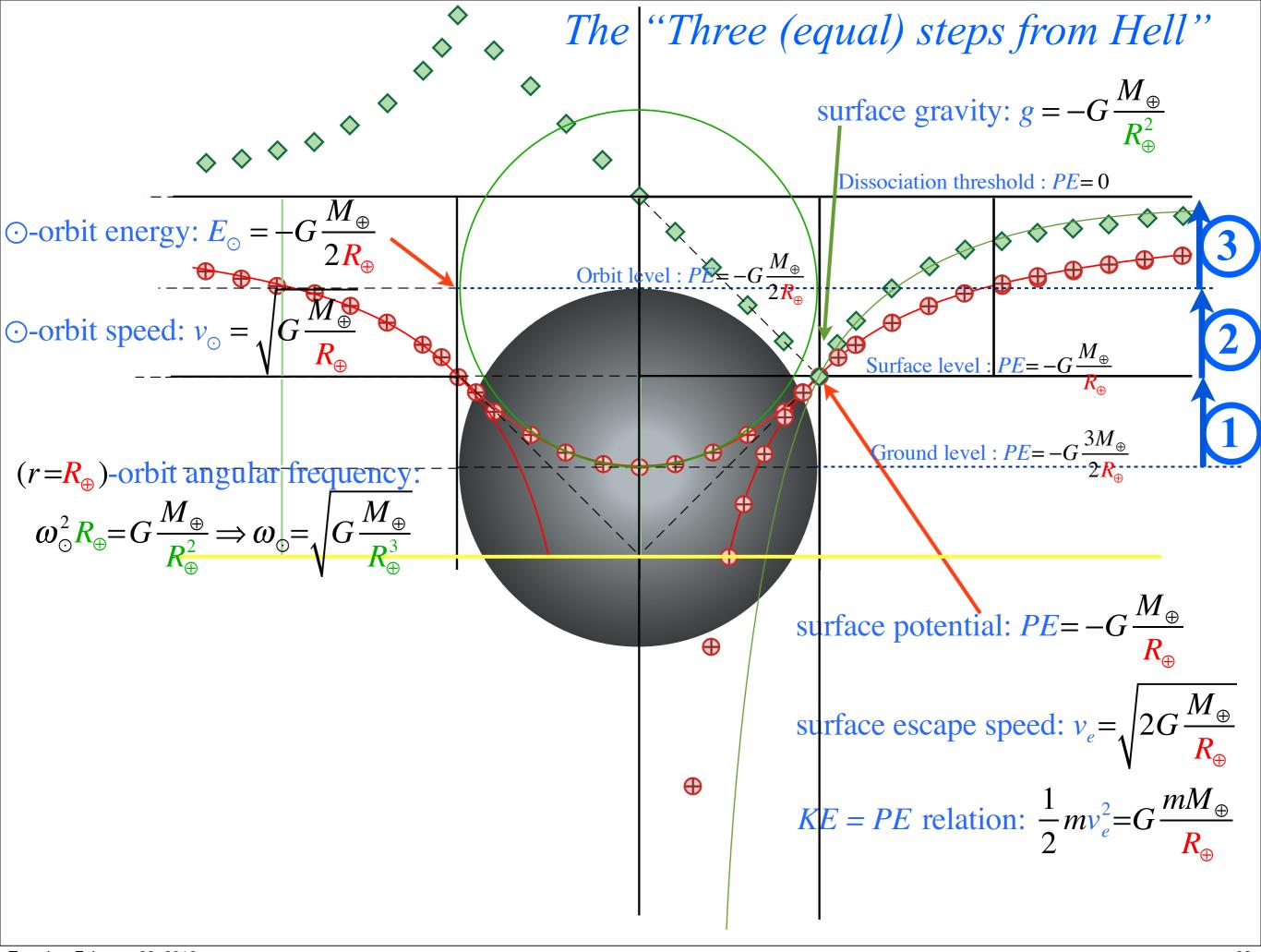


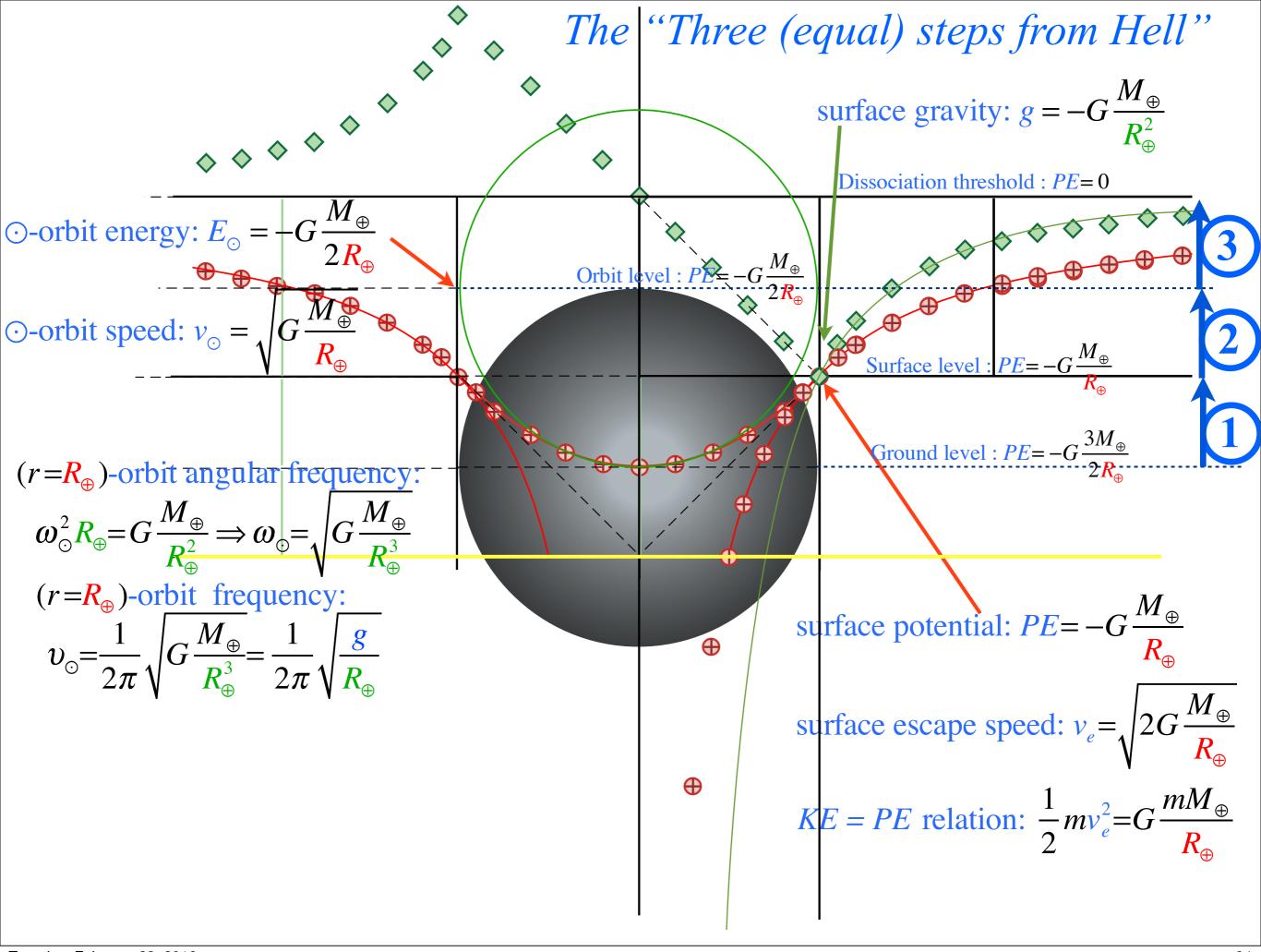


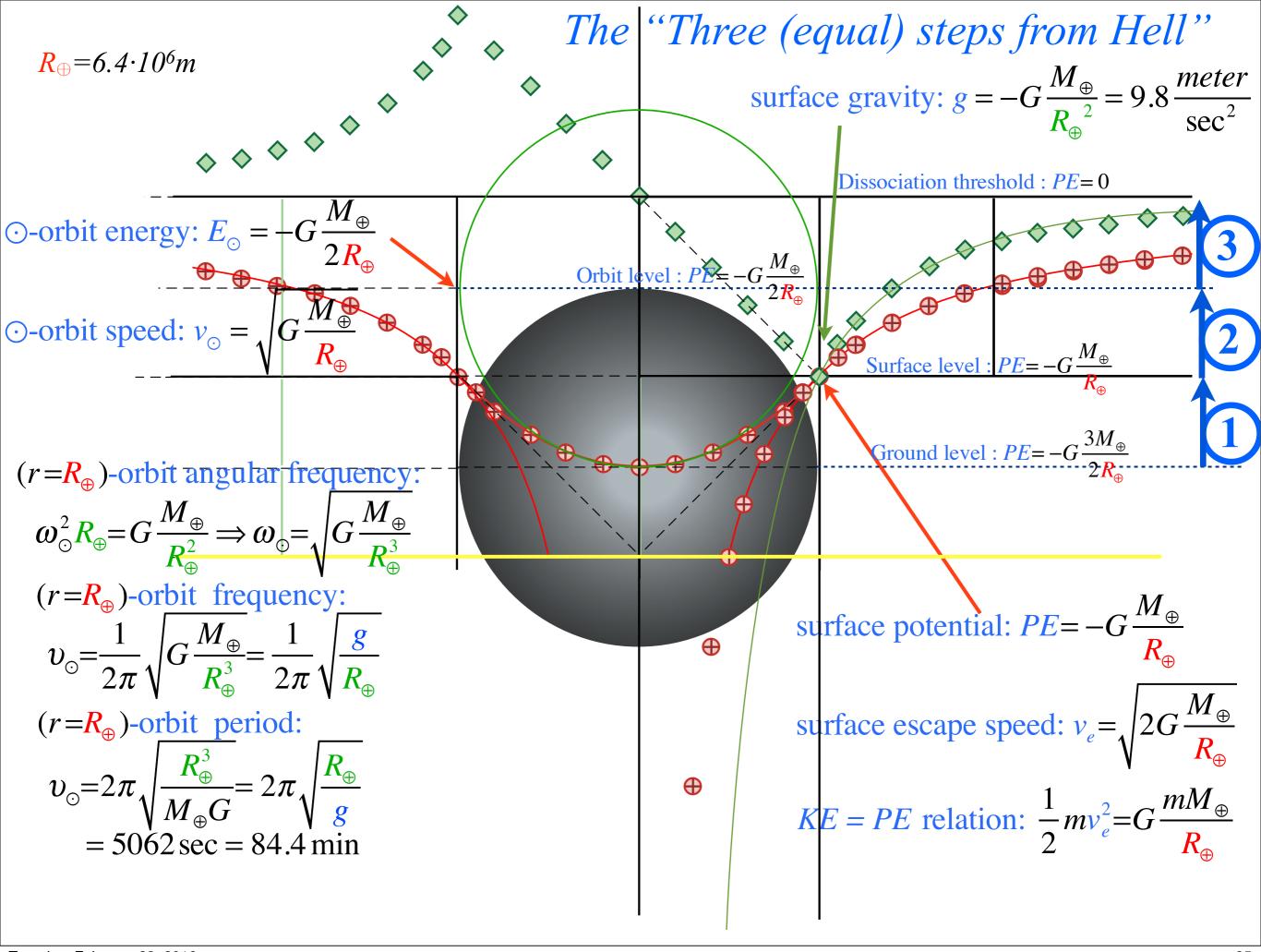


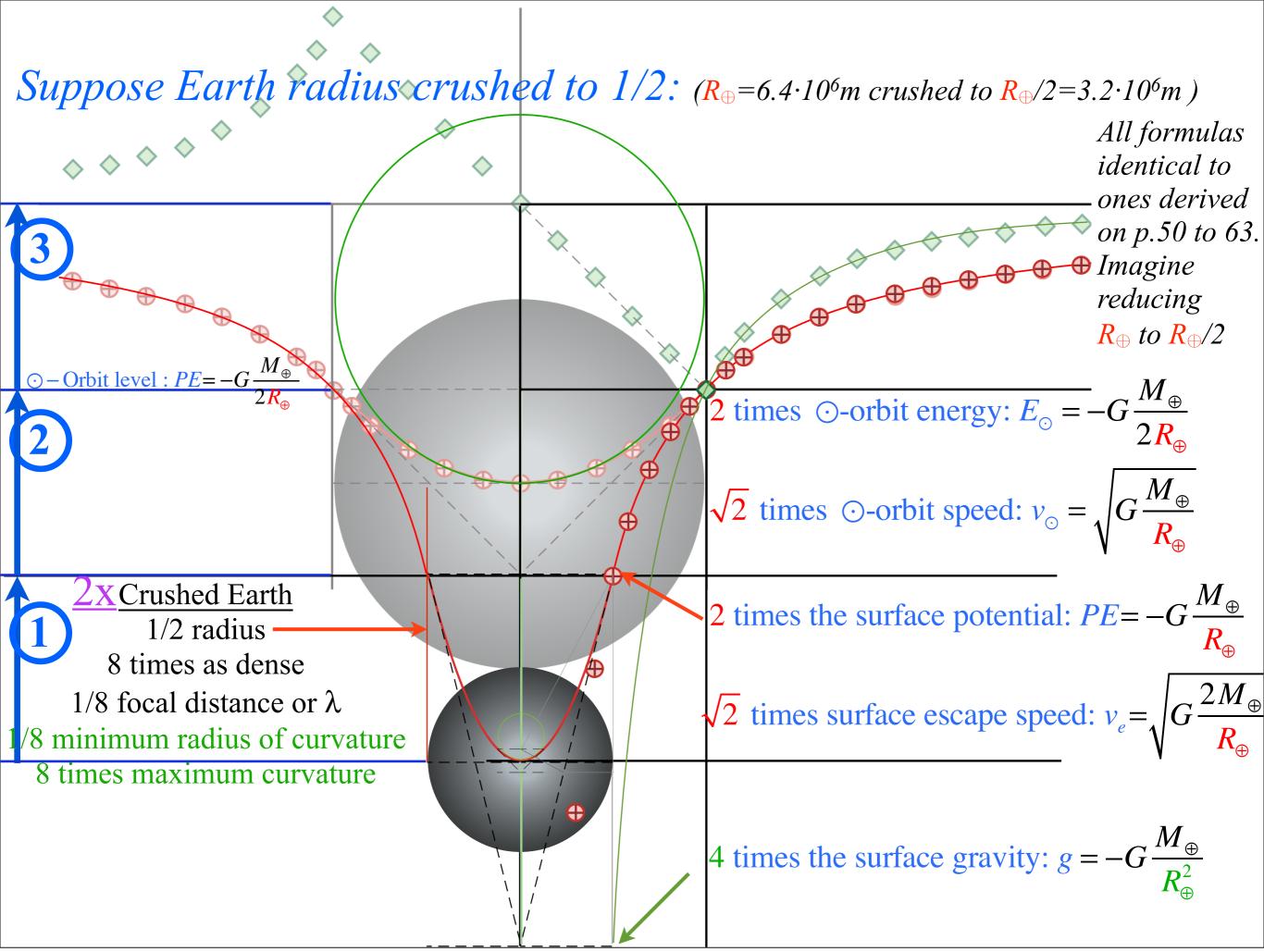












Geometry of idealized "Sophomore-physics Earth"

Coulomb field <u>outside</u> Isotropic Harmonic Oscillator (IHO) field <u>inside</u>

Contact-geometry of potential curve(s)

"Crushed-Earth" models: 3 key energy "steps" and 4 key energy "levels"

Earth matter vs nuclear matter:

Introducing the "neutron starlet" and "Black-Hole-Earth"

Earth matter Earth mass: $M_{\oplus} = 5.9722 \times 10^{24} kg. \approx 6.0 \cdot 10^{24} kg$. Density $\rho_{\oplus} = ??$

Earth radius: $R_{\oplus} = 6.371 \cdot 10^6 \, m \simeq 6.4 \cdot 10^6 \, m$ Earth volume: $(4\pi/3) R_{\oplus}^3 \simeq 4 \cdot 262 \cdot 10^{18} \sim 10^{21} \, m^3$

 $(6.4)^3 \sim 262$ and $(4\pi/3)262 = 1089 \sim 10^3$

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Density of solid Fe=7.9·10³kg/m³ Density of liquid Fe=6.9·10³kg/m³

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Nuclear matter Nucleon mass = $1.67 \cdot 10^{-27}$ kg.~ $2 \cdot 10^{-27}$ kg.

Say a nucleus of atomic weight 50 has a radius of 3 fm, or 50 nucleons each with a mass $2 \cdot 10^{-27} kg$.

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Nuclear density is $10^{-25+43} = 10^{18} kg / m^3$ or a trillion (10¹²) kilograms in the size of a fingertip (1cc).

Earth radius crushed by a factor of $0.5 \cdot 10^{-5}$ to $R_{crush \oplus} \approx 300 m$ would approach neutron-star density.

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Introducing the "Neutron starlet" 1 cm³ of nuclear matter: mass= 1012 kg.

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Introducing the "Black Hole Earth" Suppose Earth is crushed so that its

surface escape velocity is the speed of light $c \cong 3.0 \cdot 10^8 \text{m/s}$. $c \equiv 299,792,458 \text{m/s}$ (EXACTLY)

$$V_{escape} = \sqrt{(2GM/R_{\odot})}$$

(from p. 49)

 $G=6.67384(80)\cdot 10^{-11}Nm^2/C^2\sim (2/3)10^{-10}$

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$$c = \sqrt{(2GM/R_{\odot})}$$

(from p. 49)

 $G = 6.67384(80) \cdot 10^{-11} \text{Nm}^2/\text{C}^2 \sim (2/3)10^{-10}$ $R_{\odot} = 2GM/c^2 = 8.9 \text{mm} \sim 1 \text{cm}$ (fingertip size!)

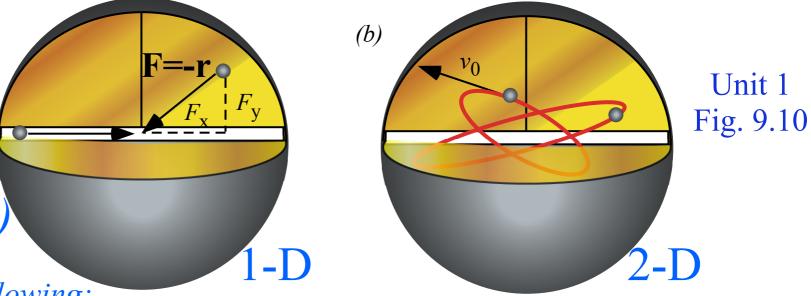
→ Introducing 2D IHO orbits and phasor geometry

Phasor "clock" geometry

I.H.O. Force law

$$F = -x$$
 (1-Dimension)

 $\mathbf{F} = -\mathbf{r}$ (2 or 3-Dimensions)



Each dimension x, y, or z obeys the following:
$$Total \ E = KE + PE = \frac{1}{2}mv^2 + U(x) = \frac{1}{2}mv^2 + \frac{1}{2}kx^2 = const.$$

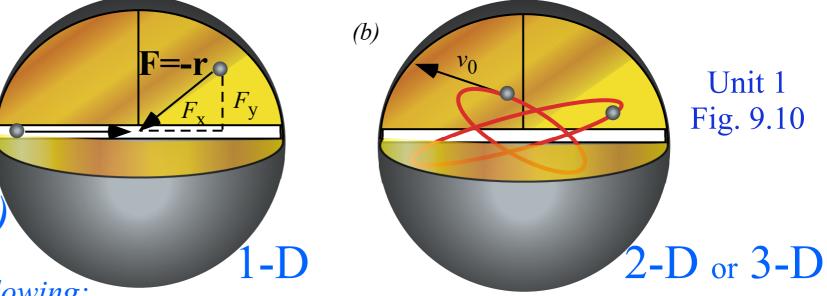
(a)

(Paths are always 2-D ellipses if viewed right!)

I.H.O. Force law

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Equations for x-motion

(a)

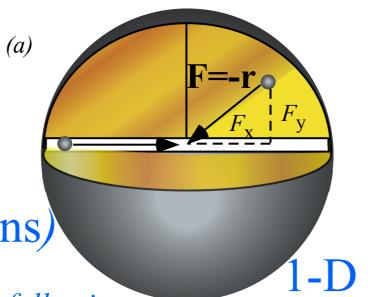
Equations for x-motion [x(t) and $v_x = v(t)$] are given first. They apply as well to dimensions [y(t) and $v_y=v(t)$] and $[z(t) \text{ and } v_z=v(t)] \text{ in the }$ ideal isotropic case.

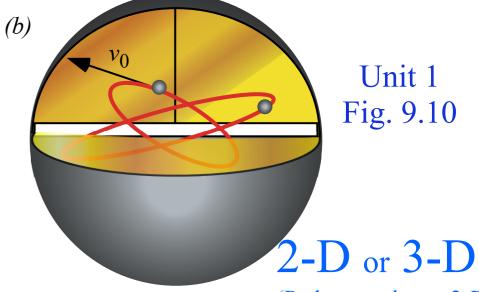
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$$1 = \frac{mv^2}{2E} + \frac{kx^2}{2E} = \left(\frac{v}{\sqrt{2E/m}}\right)^2 + \left(\frac{x}{\sqrt{2E/k}}\right)^2$$

$$1 = \frac{mv^2}{2E} + \frac{kx^2}{2E} = (\cos\theta)^2 + (\sin\theta)^2$$
Another example of the old "scale-a-circle"

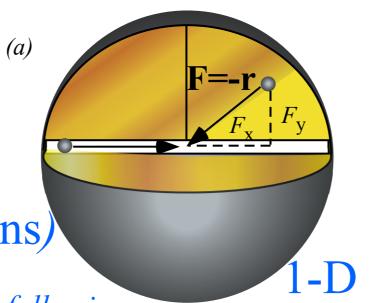
trick...

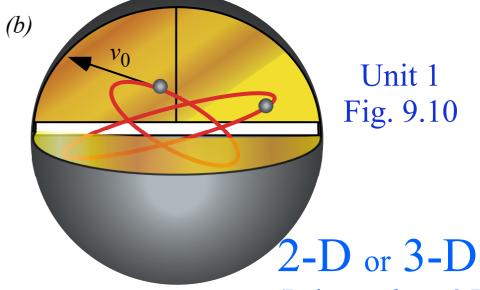
Let: (1)
$$v = \sqrt{2E/m} \cos \theta$$
, and: (2) $x = \sqrt{2E/k} \sin \theta$

I.H.O. Force law

F = -x (1-Dimension)

 $\mathbf{F} = -\mathbf{r}$ (2 or 3-Dimensions)





(Paths are always 2-D ellipses if viewed right!)

Each dimension x, y, or z obeys the following: $Total \ E = KE + PE = \frac{1}{2}mv^2 + U(x) = \frac{1}{2}mv^2 + \frac{1}{2}kx^2 = const.$

Equations for x-motion [x(t) and $v_x = v(t)$] are given first. They apply as well to dimensions [y(t) and $v_y = v(t)$] and [z(t) and $v_z = v(t)$] in the ideal isotropic case.

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trick...

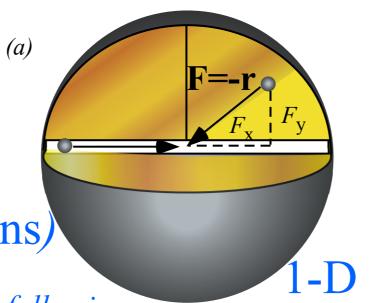
Let: (1)
$$v = \sqrt{2E/m}\cos\theta$$
, and: (2) $x = \sqrt{2E/k}\sin\theta$ def. (3) $\omega = \frac{d\theta}{dt}$

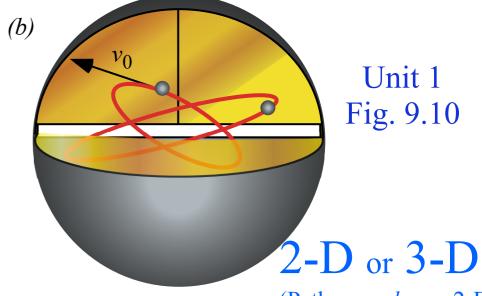
$$\sqrt{\frac{2E}{m}}\cos\theta = v = \frac{dx}{dt} = \frac{d\theta}{dt}\frac{dx}{d\theta} = \omega \frac{dx}{d\theta}$$
by (1)
by (2)
by (3)

I.H.O. Force law

$$F = -x$$
 (1-Dimension)

 $\mathbf{F} = -\mathbf{r}$ (2 or 3-Dimensions)





(Paths are always 2-D ellipses if viewed right!)

Each dimension x, y, or z obeys the following:
$$Total \ E = KE + PE = \frac{1}{2}mv^2 + U(x) = \frac{1}{2}mv^2 + \frac{1}{2}kx^2 = const.$$

Equations for x-motion [x(t) and $v_x = v(t)$] are given first. They apply as well to dimensions [y(t) and $v_y = v(t)$] and [z(t)] and $v_z=v(t)$ in the ideal isotropic case.

$$1 = \frac{mv^2}{2E} + \frac{kx^2}{2E} = \left(\frac{v}{\sqrt{2E/m}}\right)^2 + \left(\frac{x}{\sqrt{2E/k}}\right)^2$$

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Another example of the old "scale-a-circle" trick

trick...

Let: (1)
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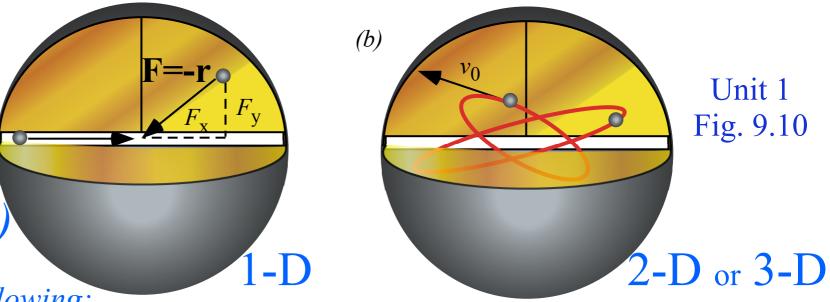
$$\frac{dy}{d\theta}\cos\theta$$

$$\frac{dy}{d$$

I.H.O. Force law

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 (1-Dimension)

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$$\frac{dy}{dt}\cos\theta = v = \frac{d\theta}{dt}\cos\theta$$

$$\frac{dy}{dt}\cos\theta = \frac{d\theta}{dt}\cos\theta$$

$$\frac{dz}{dt}\cos\theta$$

$$\frac{dz}$$

$$\omega = \frac{d\theta}{dt} = \sqrt{\frac{k}{m}}$$

$$\frac{divide \ this \ by \ (1)}{dt}$$

(Paths are always 2-D ellipses if viewed right!)

Unit 1

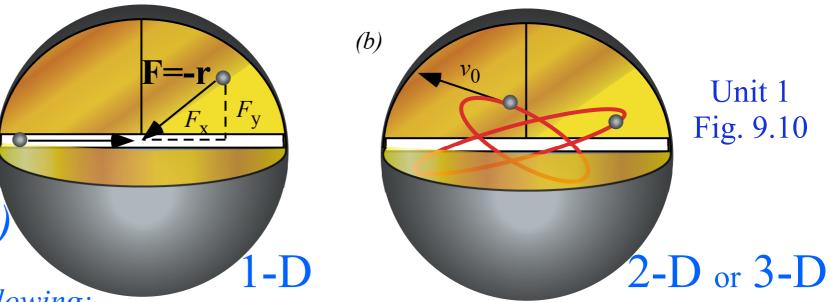
Fig. 9.10

def. (3) $\omega = \frac{d\theta}{dt}$

I.H.O. Force law

F = -x (1-Dimension)

 $\mathbf{F} = -\mathbf{r}$ (2 or 3-Dimensions)



Each dimension x, y, or z obeys the following:
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$$by (1)$$

$$by (2)$$

$$by def. (3)$$

$$\omega = \frac{d\theta}{dt} = \sqrt{\frac{k}{m}}$$

$$divide this by (1)$$

$$divide this by (1)$$

by def. (3)
$$\omega = \frac{d\theta}{dt} = \sqrt{\frac{k}{m}}$$
divide this by (1)

by integration given constant ω ?

def. (3) $\omega = \frac{d\theta}{dt}$

Unit 1

Fig. 9.10

(Paths are always 2-D

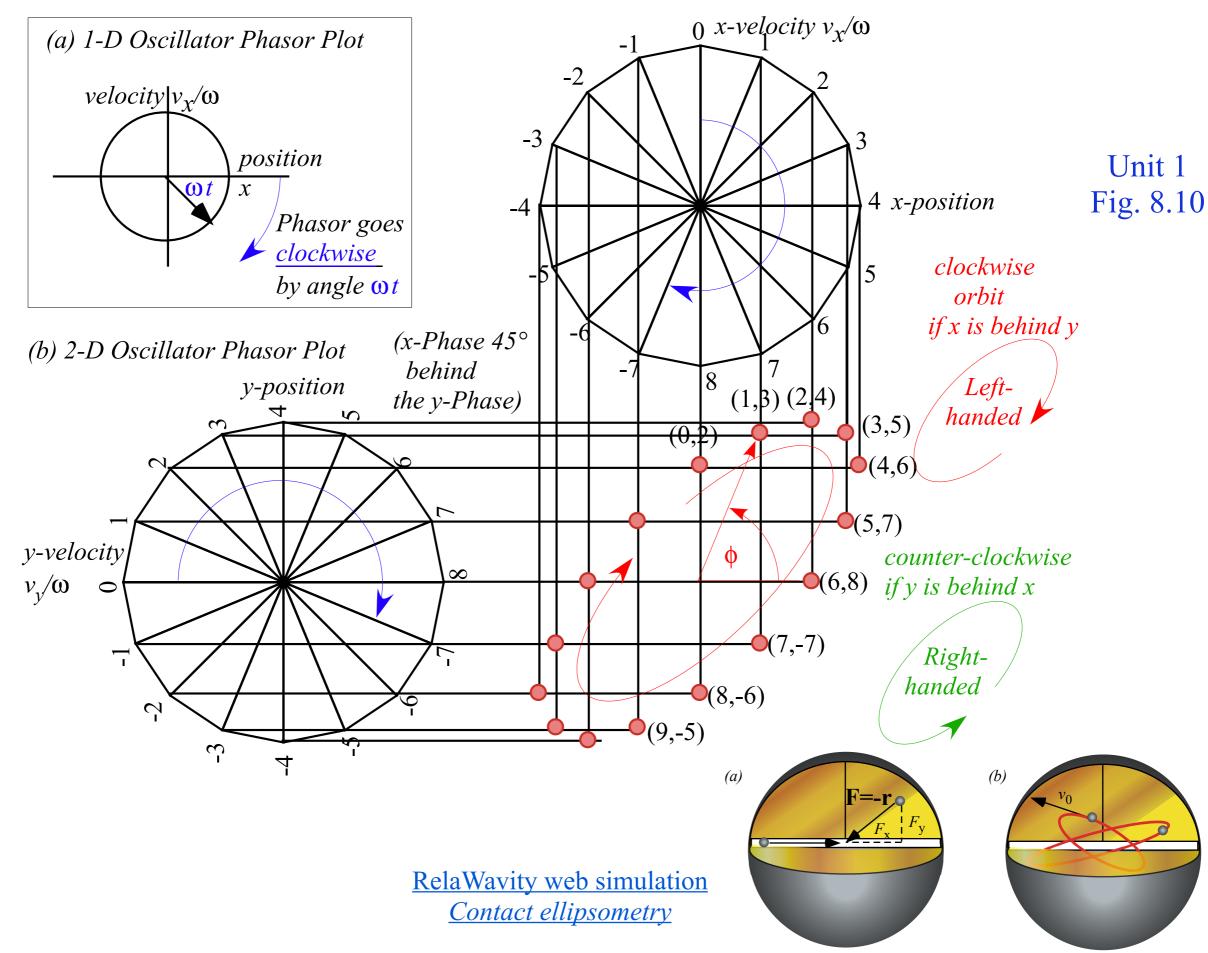
ellipses if viewed

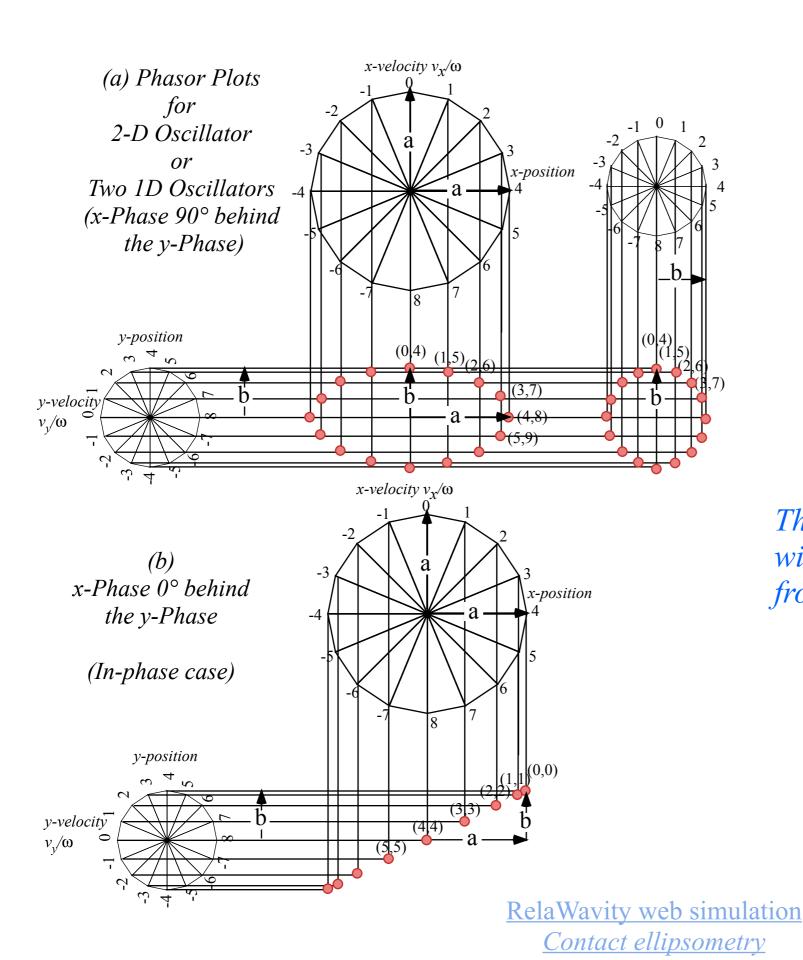
right!)



Introducing 2D IHO orbits and phasor geometry

Phasor "clock" geometry





Unit 1 Fig. 9.12

These are more generic examples with radius of x-phasor differing from that of the y-phasor.

